

Virginia Ambient Air Monitoring 2016 Data Report



Commonwealth of Virginia
Department of Environmental Quality



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This Ambient Air Monitoring Data Report is for the time period of January 1, 2016 to December 31, 2016.

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The 2016 Virginia Ambient Air Monitoring Data Report is a compilation of air pollutant measurements made by the Virginia Department of Environmental Quality, the City of Alexandria, the U.S. Department of Agriculture Forest Service, and the National Park Service. Ambient air quality was measured at 38 locations within the Commonwealth during 2016. These monitoring sites were established in accordance with EPA's siting criteria contained in 40 CFR Part 58, [Appendices D and E](#). Monitoring network operations conformed to EPA guidance documents and accepted air quality monitoring practices. All data reported for these monitoring sites were quality assured in accordance with requirements contained in 40 CFR Part 58, [Appendix A](#). Ambient concentrations of carbon monoxide, nitrogen dioxide for the hourly and annual standard, sulfur dioxide, PM10 and lead were within the EPA's national ambient air quality standards (NAAQS) in 2016. Virginia experienced a moderate ozone season in 2016. There were four days in the Richmond area, nine days in Northern Virginia, and 3 days in the Tidewater area when the ozone standard of 0.070 ppm was exceeded. For the 3-year period from 2014 through 2016 the Northern Virginia area of the Commonwealth was above the standard. Both the Richmond area and the Tidewater area are in compliance with the 0.070 ppm National Ambient Air Quality Standard (NAAQS) for ozone. For 2016 there were three exceedances of the 35 micrograms per cubic meter ($\mu\text{g}/\text{m}^3$) 24-hour standard for PM2.5 (particulate matter less than 2.5 microns): one at Bristol, one in Fairfax County and one in Arlington. The 2014 – 2016 design values for all sites in the Commonwealth for both the 24-hour and annual standard for PM2.5 are below the NAAQ standard.

Some significant changes that impacted the air quality monitoring network in 2016 are:

- Beginning in May 2016, the Office of Air Quality Monitoring began operating the Near Road Monitoring Station in Northern Virginia. The Springfield Monitor (EPA No. 51-059-0031) is required by the 2011 Nitrogen Dioxide Ambient Air Standard regulation. The station contains a nitrogen dioxide monitor, a carbon monoxide monitor and a continuous PM2.5 monitor.
- In April 2016, the lead (Pb) monitor at the NCore station at the MathScience Innovation Center was shut down. EPA approved this termination because the monitor operated for well over three years and resulted in a concentration that was less than 20% of the NAAQS.
- In April 2016, the Alexandria Colvin Street site was shut down. The site had been active since September of 2011, but was shut down due to the redundancy with the Near Road Site in Springfield.
- The PM10 monitor in King William County at West Point was shut down in June 2016.
- A manual PM2.5 FRM sampler was added to the Near Road Site in Bryan Park, City of Richmond.
- The special sampling project located at Stevenson Park in the City of Alexandria was shut down in March of 2016.

AQM is responsible for seeing that the Virginia ambient air monitoring network is maintained and operated in accordance with state and federal guidelines. Personnel from the Department of Environmental Quality (DEQ) regional offices, the City of Alexandria, the National Park Service, and the U.S. Department of Agriculture Forest Service conduct the daily operations at these sites. One of AQM's primary jobs is to support these people in their air quality monitoring efforts. AQM does this by:

- calibrating air monitoring instrumentation and associated support equipment on a set schedule
- auditing the instrumentation to insure that it is operating within set standards
- troubleshooting instrumentation problems reported by the regional operators
- supplying field operators with necessary items so they can perform their job properly
- repairing malfunctioning sampling instrumentation and ancillary equipment

Other functions:

- respond to regional and locality requests for special sampling such as emergency response or to answer citizen complaints
- coordinate efforts with the regional offices and localities to determine new air monitoring site locations
- conduct AQM generated special sampling projects to characterize a community's air quality
- furnish ambient air data to the regional offices, localities, Central Office, EPA and the EPA database
- answer FOIA requests for ambient air sampling data
- work with the regions and the localities to see that area monitoring needs are met
- work with EPA to see that necessary state and federal monitoring needs are met
- support SESARM (Southeastern States Air Resource Managers, Inc.) and MARAMA (Mid-Atlantic Regional Air Management Association of the Southeast) on routine and special projects

Criteria Pollutant Monitoring:

A portion of the air monitoring network is made up of instruments that sample for the [Criteria Pollutants](#). Sulfur dioxide, nitrogen dioxide, carbon monoxide, ozone, lead and particulate matter (PM10 & PM2.5) can injure health, harm the environment and cause property damage. EPA calls these pollutants criteria air pollutants because they have regulated them by first developing health-based criteria (science-based guidelines) as the basis for setting permissible limits. One set of limits (primary standard) protects health; another set of limits (secondary standard) is intended to prevent environmental and property damage

Special Monitoring:

In addition to overseeing the air sampling network for criteria pollutants, AQM conducts routine and short term sampling for VOCs (volatile organic compounds), Carbonyls, Toxic Metals and NOy (total reactive nitrogen). Sampled VOCs are made up of 58 HAPs (Hazardous Air Pollutants) and 56 Hydrocarbon Ozone Precursors.

1. What is the Clean Air Act?

The Clean Air Act is a federal law that provides for the protection of human health and the environment. The original Clean Air Act was passed in 1963, and the 1970 version of the law resulted in the creation of the U.S. Environmental Protection Agency (EPA), which was charged with setting and enforcing ambient air quality standards. The law was amended in 1977, and most recently in 1990. Most of the activities of the Virginia Department of Environmental Quality's Air Division come from mandates of the Clean Air Act, and are overseen by the EPA. More information on the 1990 amendments to the Clean Air Act can be found at: <http://www.epa.gov/air/caa/>.

2. What is a criteria air pollutant?

The Clean Air Act names six air pollutants that are commonly found in the air throughout the United States, and that can injure humans by causing respiratory and cardiovascular problems, and harm the environment by impairing visibility, and causing damage to animals, crops, vegetation and buildings. EPA has developed health-based criteria for these pollutants through scientific studies, and has established regulations setting permissible levels of these pollutants in the air. The "criteria" pollutants are: carbon monoxide, sulfur dioxide, nitrogen dioxide, ozone, particulate matter, and lead, and the limits that have been set for them are the [National Ambient Air Quality Standards](#) (NAAQS).

3. What is the difference between a primary and secondary National Ambient Air Quality Standard?

The National Ambient Air Quality Standards are divided into two types. The first type, the primary standard, is designed to protect human health, especially those who are most vulnerable such as children and the elderly, and people suffering from asthma, emphysema, chronic bronchitis, and heart ailments. The second type, the secondary standard, is designed to prevent damage to property and the environment. For a list of the primary and secondary National Ambient Air Quality Standards, see <https://www.epa.gov/criteria-air-pollutants/naaqs-table> or page 72 of this report.

4. How is the location of an air monitoring station decided?

Generally, the deciding factor in all Virginia air monitoring sampling is to determine where the highest pollutant concentrations will occur, and place the sampler as near as possible to that location. A wind rose is typically used to determine the prevailing wind direction for an area and identify the downwind direction from a probable source. A wind rose is a meteorological map showing the frequency and strength of winds from different directions at a specific location.

For typical criteria pollutant monitoring, the federal guidelines on siting an air monitor for measuring maximum concentrations are followed. These guidelines not only encourage siting in areas with free airflow and a minimum amount of obstructions, but they also give the height requirements for the sample inlet and the desired separation distances from obstructions such as tree lines, localized sources such as oil furnace flues, and other influences that can skew the data.

Other determining factors for placing air monitoring stations include:

- ❖ security of the site
- ❖ safety of the operator
- ❖ availability of electric power and communication service
- ❖ accessibility of the site

For more specific information, consult EPA's *Quality Assurance Handbook for Air Pollution Measurement Systems, Volume II, Section 6*, <http://www.epa.gov/ttn/amtic/qalist.html>

5. How large of an area does an air monitoring station represent?

The sampling area of a monitoring site is dependent on the parameters selected for representation, such as:

- type of pollutants being sampled
- rural vs. urban sampling
- source oriented, population oriented, or background oriented
- sampling for pollution transported from outside the Commonwealth

Many sites are also dependant on topography and meteorology of an area, which play an important role. Federal guidelines spell out the general area of representation. Some examples of varied air sampling sites are:

- A background research site in central Virginia may represent an area with a radius of 50 to 100 kilometers.
- An ozone or fine particulate site in the Shenandoah Valley may represent an elongated area with an axis running with the valley and is a hundred kilometers long but only twenty-five kilometers wide.
- A carbon monoxide sampling site in an urban street canyon setting may represent an area of only a few blocks in radius.
- A source oriented site in south central Virginia may represent an area from 0.5 to 4 kilometers in radius.

For more specific information, consult EPA's *Quality Assurance Handbook for Air Pollution Measurement Systems, Volume II, Section 6*, <http://www.epa.gov/ttn/amtic/qalist.html>

6. What is a "nonattainment" area?

A nonattainment area is a geographic area that has been determined by EPA as not meeting the air quality standards for one or more pollutants. Typically, an area is declared nonattainment based on data collected at one or more ambient air monitoring sites within the area. However, sometimes the nonattainment designation can be made based on the use of air quality models that use monitoring data from other areas. In Virginia, nonattainment areas are designated for two of the criteria pollutants, ozone and fine particulate matter (PM_{2.5}).

7. How can I find out if I live in a nonattainment area?

A list of nonattainment areas in Virginia can be found in this report on page 73. EPA has a list of all nonattainment areas in the country at <https://www.epa.gov/green-book/green-book-national-area-and-county-level-multi-pollutant-information>.

8. What are the impacts of a nonattainment designation?

To demonstrate how they plan to achieve federal air quality standards, states must draft a "State Implementation Plan," or SIP. This plan lists specific actions that the state will undertake to improve and maintain acceptable air quality, and a time frame for accomplishing these goals. The SIP may require new factories to install the newest and most effective air pollution control technologies. Other actions could be requiring older factories to retrofit their smokestacks with better pollution control devices, requiring an area to sell only reformulated gasoline during the summer months, requiring vapor recovery systems on gasoline pumps, and requiring vehicle exhaust emission checks, to name a few. SIP development is a lengthy process, and involves negotiation between the state and the EPA until it is finalized.

9. What is a Maintenance Area?

A maintenance area is an area that has formerly been designated nonattainment, but is now recognized by EPA as meeting the NAAQS. A maintenance area must have an approved "maintenance plan" to meet and maintain air quality standards.

10. What is a design value?

A design value is a statistic that describes the air quality status of a given area relative to the level of the National Ambient Air Quality Standards (NAAQS). Design values are typically used to classify nonattainment areas, assess progress towards meeting the NAAQS, and develop control strategies. Design values are expressed as concentrations in the ambient air and are calculated according to regulatory specifications to determine the highest monitored concentration in an attainment or non-attainment area.

11. How can I get current or historical air quality data?

Current ozone data for Virginia, as well as current AQI and air quality forecasts can be obtained at http://vadeq.tx.sutron.com/cgi-bin/aqi_rpt.pl. Summary air quality data PM2.5 can also be found at <http://www.deq.virginia.gov/Programs/Air/AirMonitoring/ParticulateMonitoring.aspx>. Annual monitoring data reports can be found at <http://www.deq.virginia.gov/Programs/Air/AirMonitoring/Publications.aspx>. EPA provides monitoring data, as well as maps, on the web at <https://www.epa.gov/outdoor-air-quality-data> and www.epa.gov/air/emissions. Detailed data for monitoring sites in Virginia can also be obtained by contacting the VA DEQ Office of Air Quality Monitoring, or from EPA's AQS Data Mart at https://aqs.epa.gov/aqsweb/documents/data_mart_welcome.html.

12. What do I do if I have a complaint about air quality in my neighborhood?

Contact the DEQ regional office in your area. To see a list of regional offices and phone numbers, see page 63 of this report, or visit <http://www.deq.virginia.gov/Programs/PollutionResponsePreparedness.aspx>.

13. Who can I call about an indoor air quality problem, such as mold or radon gas?

Your local health department may be able to assist you with some indoor air quality problems. See <http://www.vdh.virginia.gov/local-health-districts> for the health department office in your area or the Division of Environmental Epidemiology at (804) 864-8182. Other excellent sources of information on indoor air quality can be found on EPA's website at www.epa.gov/iaq/index.html and through the American Lung Association website at www.lungusa.org.

Criteria Pollutants

PM_{2.5} is particulate matter (PM) that is less than or equal to 2.5 micrometers (a micrometer is one millionth of a meter) in aerodynamic diameter. These particles are often called “fine particles” because of their small size. Fine particles originate from a variety of man-made stationary and mobile sources, such as factory smoke stacks and diesel engines, as well as from natural sources, such as forest fires and dust storms. These particles may be emitted directly into the air, or they may be formed by chemical reaction in the atmosphere from gaseous emissions of sulfur dioxide (SO₂), nitrogen oxides (NO_x), and volatile organic compounds (VOCs) and other chemicals.

Scientific research has linked fine particle pollution to human health problems. The particles are easily inhaled deep into the lungs, and can actually enter the bloodstream. Particle pollution is of particular concern to people with heart or lung disease, such as coronary artery disease, congestive heart failure, asthma, or chronic obstructive pulmonary disease (COPD). Older adults are at risk because they may have underlying, undiagnosed heart or lung problems. Young children are also at risk because their lungs are still developing, they are more likely to have asthma or acute respiratory disease, and they tend to spend longer periods of time at high activity levels, causing them to inhale more particles than someone at rest. Even otherwise healthy people may suffer short-term symptoms such as eye, nose, throat irritation, coughing, and shortness of breath during episodes of high particulate levels.

PM_{2.5} air quality standards were implemented by EPA in 1997 to protect against the health effects of fine particle pollution. On December 14, 2012, EPA strengthened the nation’s air quality standards for fine particle pollution to improve public health protection by revising the primary annual PM_{2.5} standard to 12.0 µg/m³ and retaining the 24-hour fine particle standard of 35 µg/m³. For more information, see http://www.epa.gov/sites/production/files/2016-04/documents/overview_factsheet.pdf.

In addition to health problems, fine particle pollution contributes to haze that causes deterioration of visibility in scenic areas, and also deposits harmful compounds on the soil and water. Unlike ozone, which is a seasonal pollutant in most areas of the country, particle pollution can occur year-round, and is monitored throughout the year in Virginia. The Virginia DEQ PM_{2.5} monitoring network uses three different types of samplers to monitor fine particulate in the state:

PM_{2.5} 24-hour Mass Sampler: This Federal Reference Method (FRM) sampler collects particulate matter on a stretched Teflon filter media. Four samplers (Henrico Co., Vinton, Virginia Beach, and Fairfax Co.) collect 24-hour samples every day. The rest of these samplers collect 24-hour samples on a one-in-three day schedule. The 3-day monitoring schedule can be found at <http://www3.epa.gov/ttn/amtic/calendar.html>. Filters are retrieved from the field and shipped via courier to the Virginia Division of Consolidated Laboratory Services (DCLS) in Richmond. At the laboratory, the filters are equilibrated for a minimum of 24 hours prior to the final weighing.

PM_{2.5} 24-hour Speciation: Speciated PM_{2.5} data are collected at one site in Virginia, the MathScience Innovation Center in Henrico Co., using two co-located samplers that operate simultaneously. One sampler, the MetOne SASS, collects particulate matter on two filters, one nylon and the other Teflon. The second sampler, the URG 3000N carbon sampler, uses a quartz filter to collect particulate matter. The samplers run for 24 hours, on a one-in-three day sampling schedule. After the completion of a sample run, the instrument operator removes the exposed filters and ships them via refrigerated container to an EPA contract lab, where the filters are analyzed for the following:

- Teflon filter: thirty-three trace elements including aluminum, antimony, arsenic, bromine, calcium, iron, lead, silicon, titanium, vanadium, and zirconium
- Nylon filter: cations (ammonium, potassium, sodium) and anions (nitrate, sulfate)
- Quartz filter: carbons (carbonate carbon, elemental carbon, organic carbon, total carbon)

The resulting data provide a “chemical fingerprint” of air masses moving through the Richmond area. These data, in conjunction with historical data from other speciation sites, including those outside Virginia, give a representative picture of the constituents of the air samples, which help identify sources of high values and show how the air masses move over a broad area.

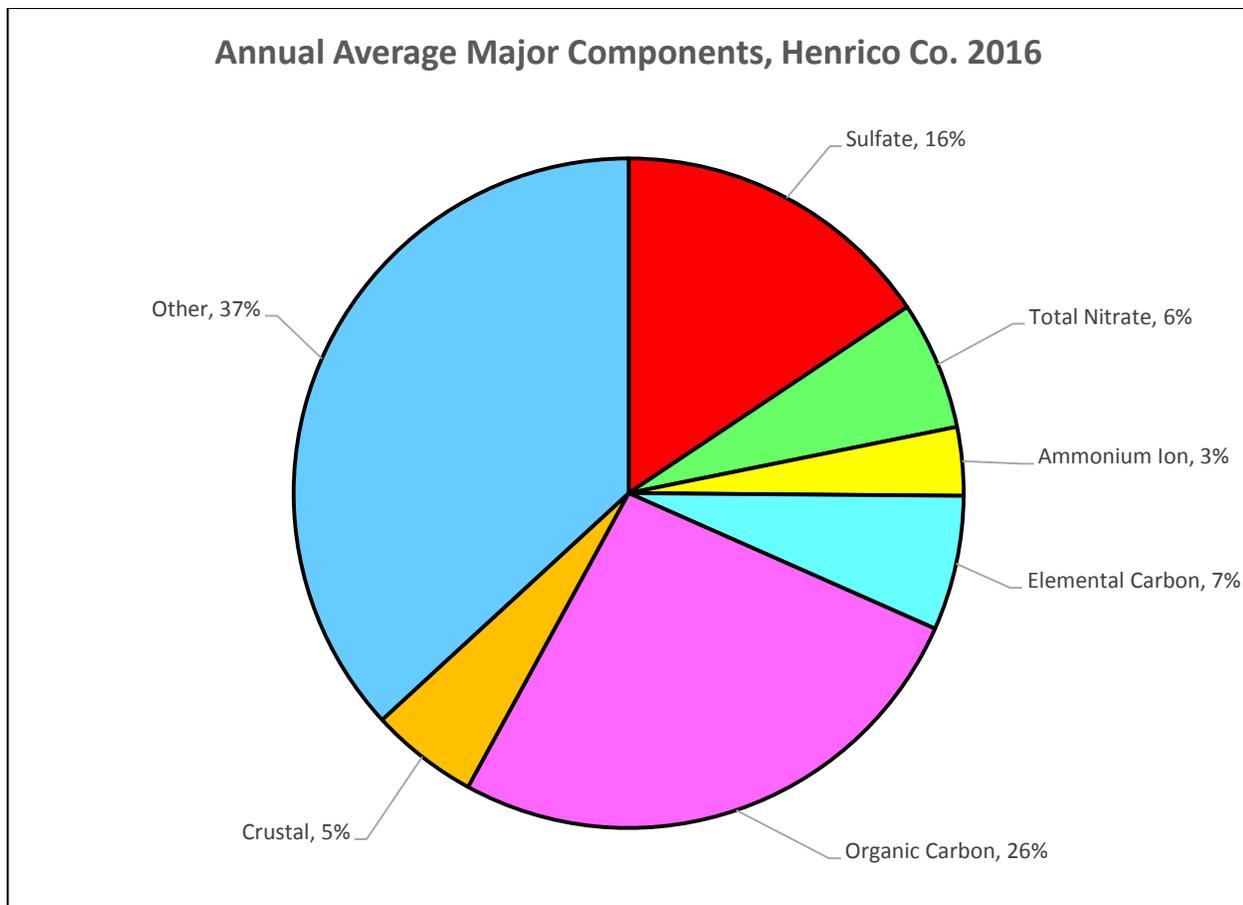
PM_{2.5} Continuous: Unlike the PM_{2.5} 24-hour sampler, these instruments collect particulate samples on a continuous basis, and data can be compiled into hourly and 24-hour averages. PM_{2.5} continuous samplers are operated in Henrico Co., Roanoke Co., Fairfax Co., Shenandoah National Park, Frederick Co., Albemarle Co., and the Cities of Richmond and Hampton. Some of these continuous monitors are designated as Federal Equivalent Method (FEM) monitors and others are used strictly for forecasting purposes only.

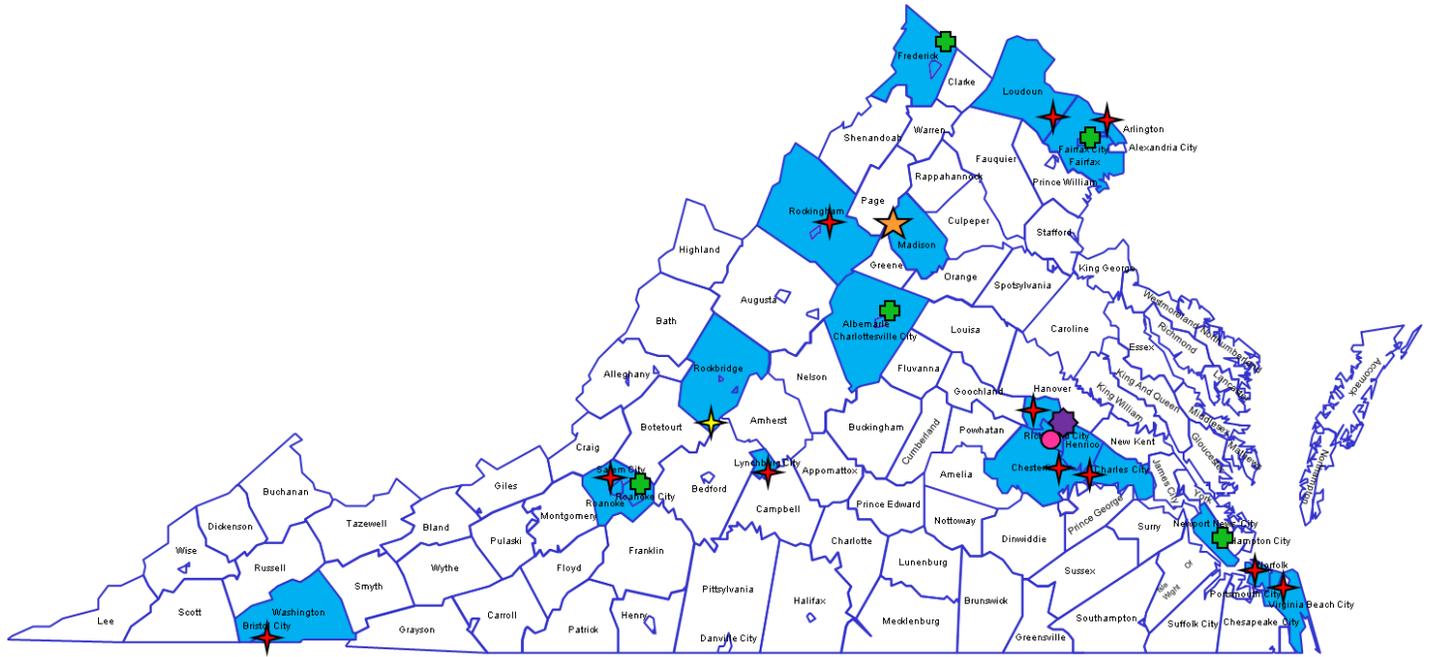
The purpose of each of the PM_{2.5} samplers is determined by the type and designation of the sampler. The FRM and the FEM samplers collect data that are used to determine if the state is complying with the national ambient air quality standards (NAAQS) for particulate matter. The speciation sampler collects data about the composition of particulate matter in Virginia, and is useful for identifying potential sources of air pollution both within and outside the state boundaries. The FRM and speciation monitors are manual, filter-based methods, and the samples they collect must be transported to a laboratory for processing. Consequently, they are not useful for reporting real-time air quality conditions. Some of the continuous particulate monitors provide real-time data on fine particulate levels. The hourly average data are polled by a central computer, and then posted on the agency website at http://vadeq.tx.sutron.com/cgi-bin/select_curlev.pl?user_param=88502. The data are also simultaneously sent to EPA’s national air quality website at www.airnow.gov.

In addition to the PM_{2.5} network operated by the DEQ, the National Park Service and the USDA Forest Service operate PM_{2.5} samplers at Big Meadows in Shenandoah National Park, and in Rockbridge Co. as part of the IMPROVE (Interagency Monitoring of Protected Visual Environments) network. This network employs different sampling methods than those used by the DEQ. Data for the IMPROVE network can be found on the internet at <http://vista.cira.colostate.edu/improve>.

Some of the continuous monitors are designated as Federal Equivalent Method (FEM) monitors, meaning they may be used to determine NAAQS compliance as well as to report the AQI, whereas the TEOM is used strictly for AQI reporting and forecasting purposes.

PM_{2.5} Monitoring Network





-  FRM Sampler
-  IMPROVE sampler
-  FRM and Continuous Samplers
-  FRM, Speciation, Continuous Sampler
-  Continuous & IMPROVE sampler, Big Meadows, NPS
-  FEM Continuous Sampler

National Ambient Air Quality Standards (NAAQS)

Primary Standard for PM_{2.5}:

- Annual Arithmetic Mean – the 3 year average of the weighted annual mean PM_{2.5} concentration must not exceed 12.0 µg/m³.
- 24-Hour concentration – the 3 year average of the 98th percentile of 24-hour concentrations must not exceed 35 µg/m³.

Secondary Standard for PM_{2.5}:

- Annual Arithmetic Mean – the 3 year average of the weighted annual mean PM_{2.5} concentration must not exceed 15.0 µg/m³.

2014-2016 PM_{2.5} 24-hour Averages, 98th Percentile Values (µg/m³, LC)				
Site	2014	2015	2016	3-Year Average
(101-E) Bristol	15.1	13.4	24.9**	17.8
(26-F) Rockingham Co.	21.6	22.7	17.6	20.6
(28-J) Frederick Co.	24.2	23.0	17.5	21.5
(33-A) Albemarle Co.	15.1	17.3	14.2	15.5
(19-A6) Roanoke Co.	16.6	18.0	15.0*	16.5
(110-C) Salem	16.8	18.8	15.0	16.8
(155-Q) Lynchburg	16.5	15.0	12.8	14.7
(71-D) Chesterfield Co.	16.0	16.8	15.2	16.0
(72-M) Henrico Co.	15.7	17.2	14.8	15.9
(72-N) Henrico Co.	16.0	15.6	15.0	15.5
(75-B) Charles City Co.	16.0	17.2	13.7	15.6
(158-X) Richmond	--	21.9	17.6	NA
(179-K) Hampton	15.3	16.8	12.5	14.8
(181-A1) Norfolk	15.6	17.3	13.3	15.4
(184-J) Va. Beach	19.8	18.0	14.6	17.4
(38-I) Loudoun Co.	19.2	21.9	15.6	18.9
(47-T) Arlington Co.	19.2	20.9	17.9	19.3
(46-B9) Franconia, Fairfax Co.	18.0	19.7	16.9	18.2

* Annual value did not meet completeness criteria

** Influenced by fires in Tennessee, Virginia and Western NC

NAAQSPrimary Standard for PM_{2.5}:

- Annual Arithmetic Mean – the 3 year average of the weighted annual mean PM_{2.5} concentration must not exceed 12.0 µg/m³.
- 24-Hour concentration – the 3 year average of the 98th percentile of 24-hour concentrations must not exceed 35 µg/m³.

Secondary Standard for PM_{2.5}:

- Annual Arithmetic Mean – the 3 year average of the weighted annual mean PM_{2.5} concentration must not exceed 15.0 µg/m³.

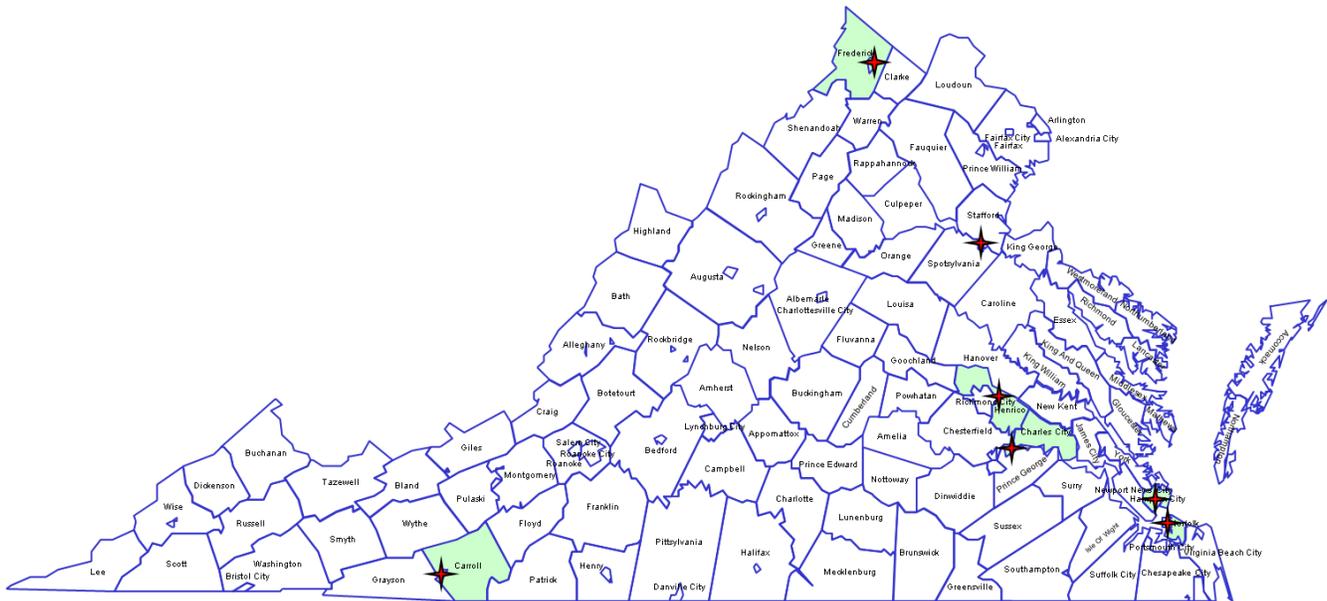
2014-2016 PM_{2.5} Weighted Annual Arithmetic Means (µg/m³, LC)				
Site	2014	2015	2016	3-Year Average
(101-E) Bristol	8.6	7.4	8.0	8.0
(26-F) Rockingham Co.	8.7	8.3	7.4	8.1
(28-J) Frederick Co.	9.0	9.0	7.4	8.5
(33-A) Albemarle Co.	7.4	7.2	6.8	7.1
(19-A6) Roanoke Co.	8.1	7.7	6.7*	7.5
(110-C) Salem	8.5	8.5	7.6	8.2
(155-Q) Lynchburg	7.6	7.1	6.8	7.2
(71-D) Chesterfield Co.	8.4	8.3	7.4	8.0
(72-M) Henrico Co.	8.1	7.7	7.0	7.6
(72-N) Henrico Co.	8.2	7.7	7.0	7.6
(75-B) Charles City Co.	7.9	7.5	6.5	7.3
(158-X) Richmond	--	10.0	8.5	NA
(179-K) Hampton	7.6	7.2	6.1	6.9
(181-A1) Norfolk	8.0	7.7	6.6	7.5
(184-J) Va. Beach	8.1	7.8	6.7	7.5
(38-I) Loudoun Co.	8.5	9.0	7.1	8.2
(47-T) Arlington Co.	8.7	9.1	7.5	8.5
(46-B9) Franconia, Fairfax Co.	8.2	8.0	6.7	7.6

* Annual value did not meet completeness criteria.

PM₁₀ is particulate matter comprised of solid particles or liquid droplets with an aerodynamic diameter of less than or equal to 10 micrometers, and is sometimes referred to as “coarse particles.” PM₁₀ particles are larger than PM_{2.5}, but are still in a size range that can pose health problems because they can be inhaled, and retained in the human respiratory system, causing breathing difficulties, and eye, nose and throat irritation. In addition to the health effects of PM₁₀, these particles can impair visibility, can contribute to climate change, and result in “acidic dry deposition.” Acidic dry deposition occurs when particles containing acidic compounds fall to the ground. The acidic particles can corrode surfaces that they settle on, and can increase the acidity of the soil and water.

The National Ambient Air Quality Standards, or NAAQS, for particulate matter were revised in September 2006. EPA changed the existing standards for PM₁₀ by revoking the annual standard of 50 micrograms per cubic meter, because current scientific evidence did not support a link between long-term exposure to coarse particles and health problems. However, the 24-hour PM₁₀ standard was retained to protect citizens from effects of short-term exposures. For additional information on the revised particulate matter standards, see <https://www.epa.gov/pm-pollution>.

To measure PM₁₀, ambient air is drawn into a sampler that uses a particle size discrimination inlet. The inlet is designed so that particles in the size range of 10 micrometers (also called microns) or below stay suspended in the air stream, while larger particles settle out. The sample air flows across an 8 x 10 inch micro-quartz filter at a rate of 40 cubic feet per minute for a 24-hour period. The particles are captured on the filter, which is weighed before and after sampling, and the PM₁₀ concentration is determined by dividing the change in filter mass by the volume of sampled air. The resulting PM₁₀ concentration is reported as micrograms per cubic meter ($\mu\text{g}/\text{m}^3$). The filters are processed at the DEQ Office of Air Quality Monitoring. The normal sampling schedule is once every sixth day from midnight to midnight. The 6-day monitoring schedule can be found at <http://www3.epa.gov/ttn/amtic/calendar.html>.



VA Department of Environmental Quality

PM10 Monitoring Sites

National Ambient Air Quality Standards (NAAQS)

Primary Standard for PM₁₀:

- 24- hour concentration not to exceed 150 µg/m³ more than once per year averaged over three years. An exceedance means a 24-hour average value that is above the level of the 24-hour standard after rounding to the nearest 10 µg/m³.

Secondary Standard for PM₁₀:

- Same as Primary.

2014-2016 PM ₁₀ 24-Hour Average Concentrations (units in µg/m ³ STD)							
Site	2014		2015		2016		>150 µg/m ³
	1 st Max	2 nd Max	1 st Max	2 nd Max	1 st Max	2 nd Max	
(23-A) Carroll Co.	19	19	38	26	32	31	0
(134-C) Winchester	19	17	33	23	27	24	0
(72-M) Henrico Co.	26	23	40	27	25	24	0
(154-M) Hopewell	26	22	25	22	24	23	0
(179-K) Hampton	25	18	27	26	25	24	0
(181-A1) Norfolk	45**	23	24	24	25	22	0
(130-E) Fredericksburg	26	21	29	27	22	21	0

** Max influenced by construction activity

[Carbon monoxide](#) (CO) is a colorless, odorless gas that is produced by incomplete burning of carbon compounds in fossil fuels (gasoline, natural gas, coal, oil, etc.). Over half of the CO emissions in the country come from motor vehicle exhaust. Other sources include construction equipment, boats, lawnmowers, woodstoves, forest fires, and industrial manufacturing processes.

CO concentrations are higher in the vicinity of heavily traveled highways, and drop rapidly the further the distance from the road. Ambient levels of carbon monoxide tend to be higher in the colder months due to “thermal inversions” that trap pollutants close to the ground. A thermal inversion occurs when the temperature of the air next to the ground is colder than air above it. When this happens, the air resists vertical mixing that can help the pollutants to disperse, forming a layer of smog close to the ground.

Carbon monoxide is harmful because it reacts in the bloodstream, reducing the amount of oxygen that is supplied to the heart and brain. CO can be harmful at lower levels to people who suffer from cardiovascular disease, like angina, arteriosclerosis, or congestive heart failure. At high levels, CO can impair brain function, causing vision problems, reduce manual dexterity, and reduce ability to perform complicated tasks. At very high levels, carbon monoxide can be deadly.

Carbon monoxide in the ambient air is measured continuously with an electronic instrument that uses NDIR, “non-dispersive infrared” photometry. The instrument has a pump that continuously draws air through a sample chamber that contains an infrared light source and a detector. Any CO molecules that are present in the sample air absorb some of the infrared light, reducing the intensity of the light reaching the detector. The portion of the infrared light absorbed by the CO molecules is converted into an electrical signal corresponding to the CO concentration, and stored in the instrument computer.



National Ambient Air Quality Standards (NAAQS)

Primary Standard for CO:

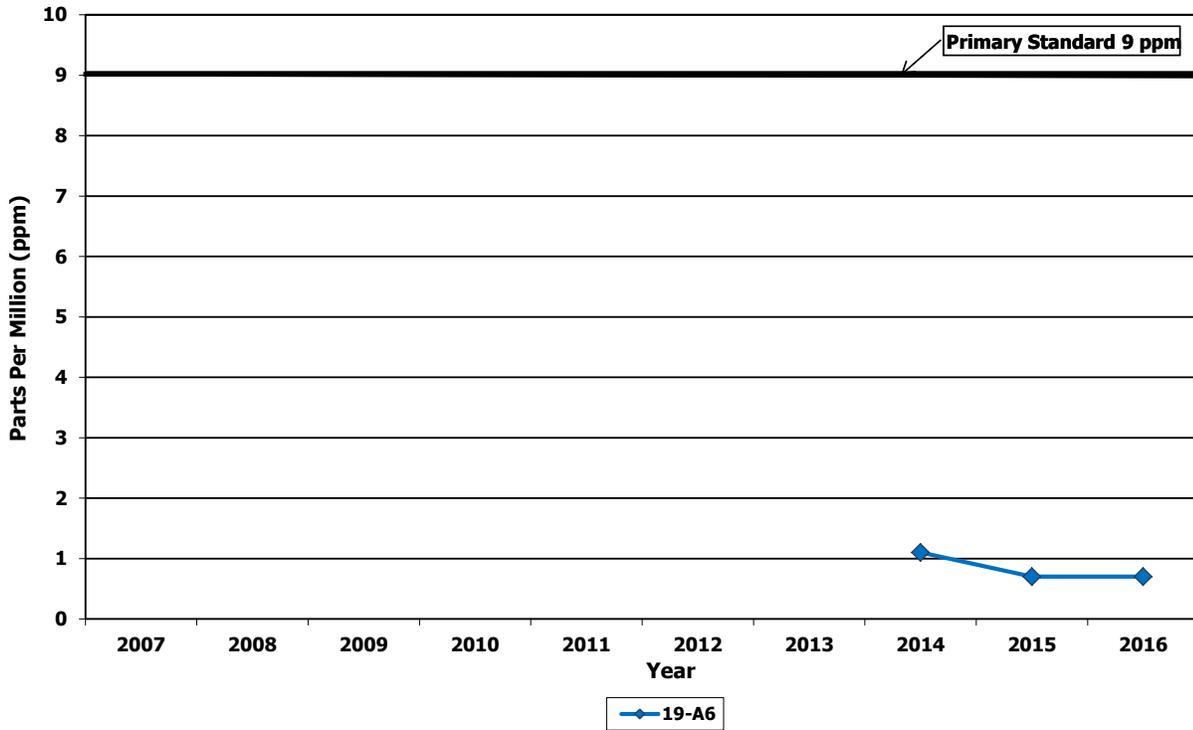
- 8-hour average not to exceed 9 ppm more than once per year.
- 1-hour average not to exceed 35 ppm more than once per year.

There are no Secondary Standards for CO because it does not harm vegetation or buildings.

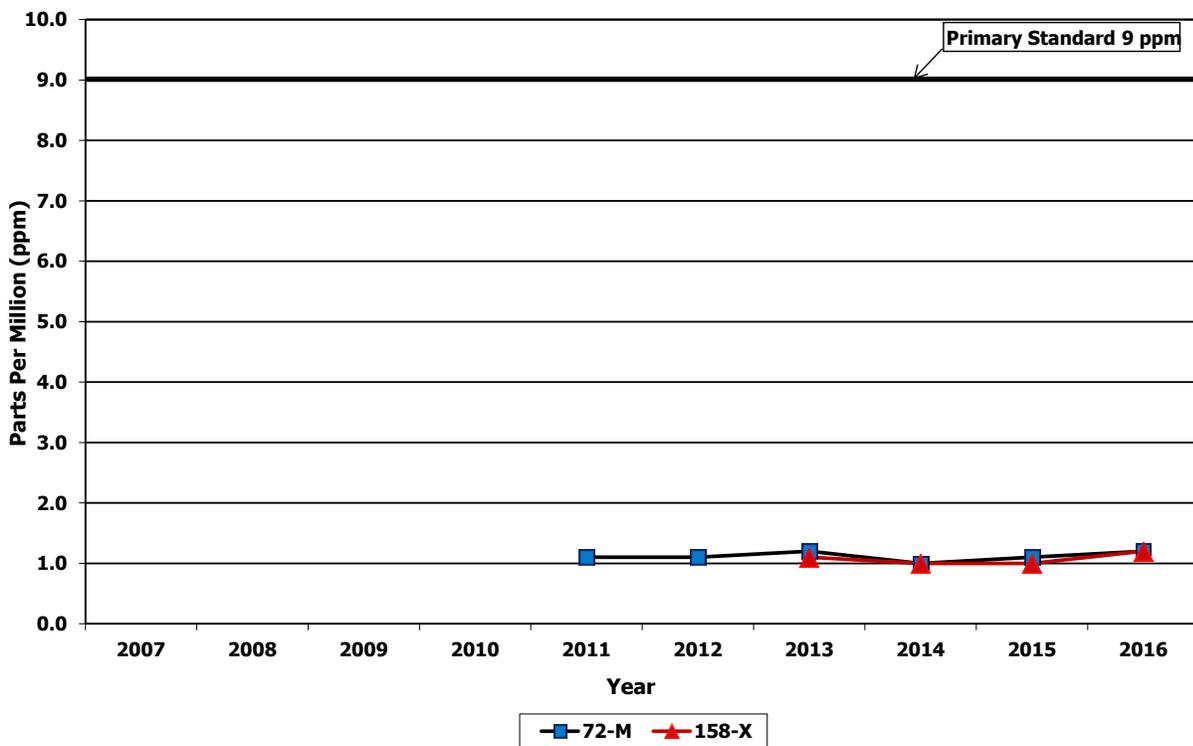
Site	2016			
	1-Hour Avg. (ppm)		8-Hour Avg. (ppm)	
	1 st Max.	2 nd Max.	1 st Max.	2 nd Max.
(19-A6) Roanoke Co.	1.1	1.1	.9	.7
(72-M) Henrico Co.	1.5	1.4	1.3	1.2
(158-X) Richmond	1.8	1.7	1.6	1.2
(179-K) Hampton	1.0	1.0	.7	.7
(181-A1) Norfolk	1.2	1.2	.8	.7
(47-T) Arlington Co.	3.7	3.7	1.8	1.6

Eight Hour Averages stated as Ending Hour

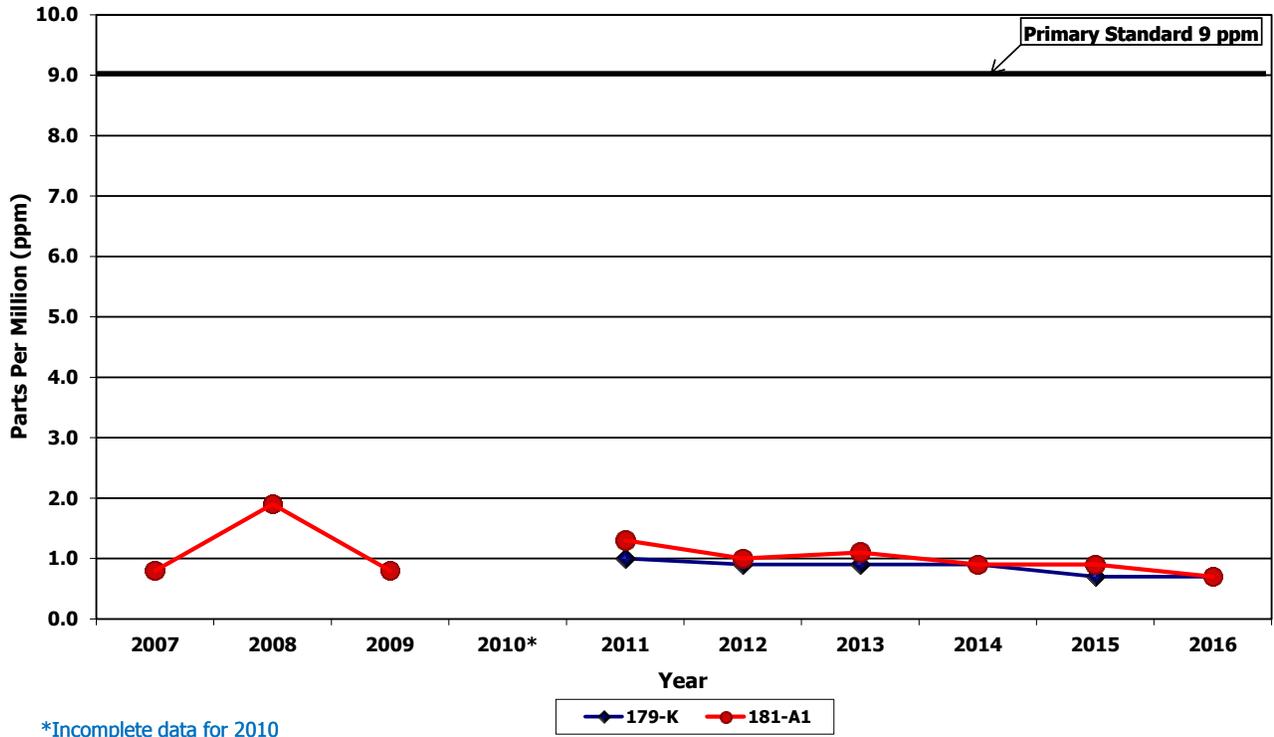
Carbon Monoxide - Blue Ridge Region Eight Hour 2nd Maximum



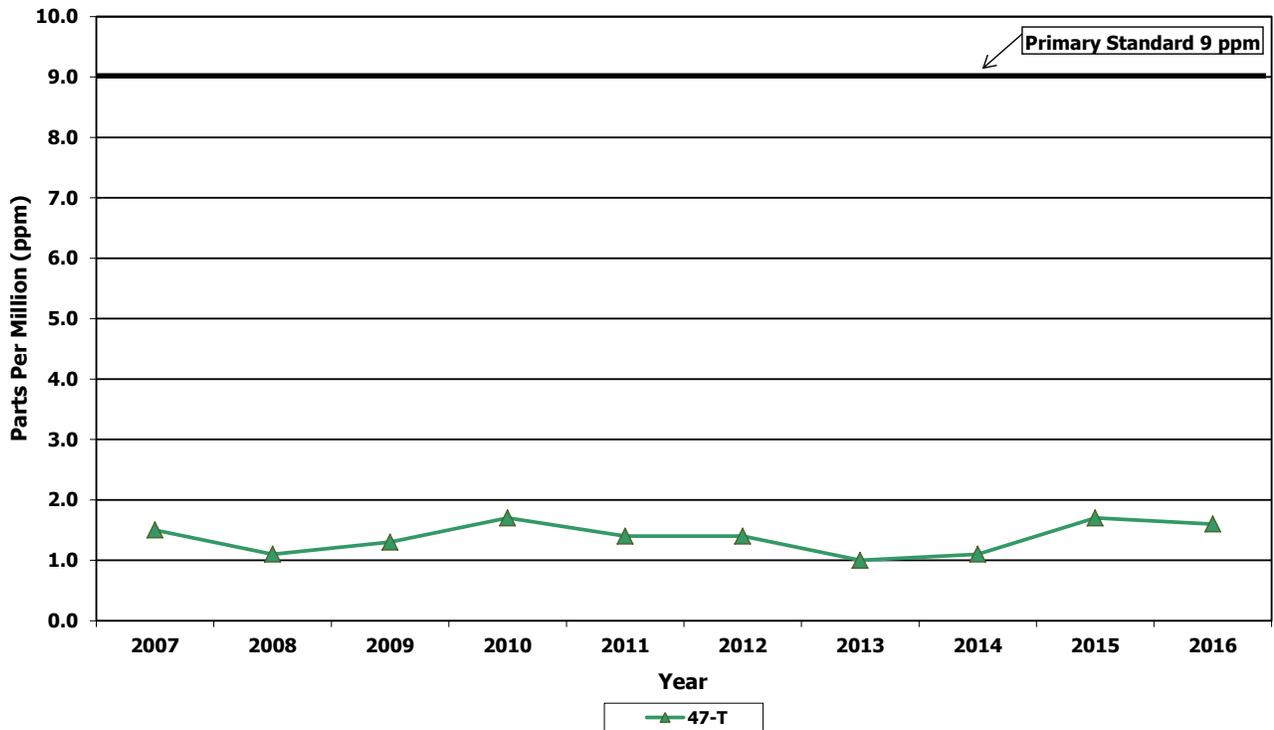
Carbon Monoxide - Piedmont Region Eight Hour 2nd Maximum



Carbon Monoxide - Tidewater Region Eight Hour 2nd Maximum



Carbon Monoxide - Northern Region Eight Hour 2nd Maximum

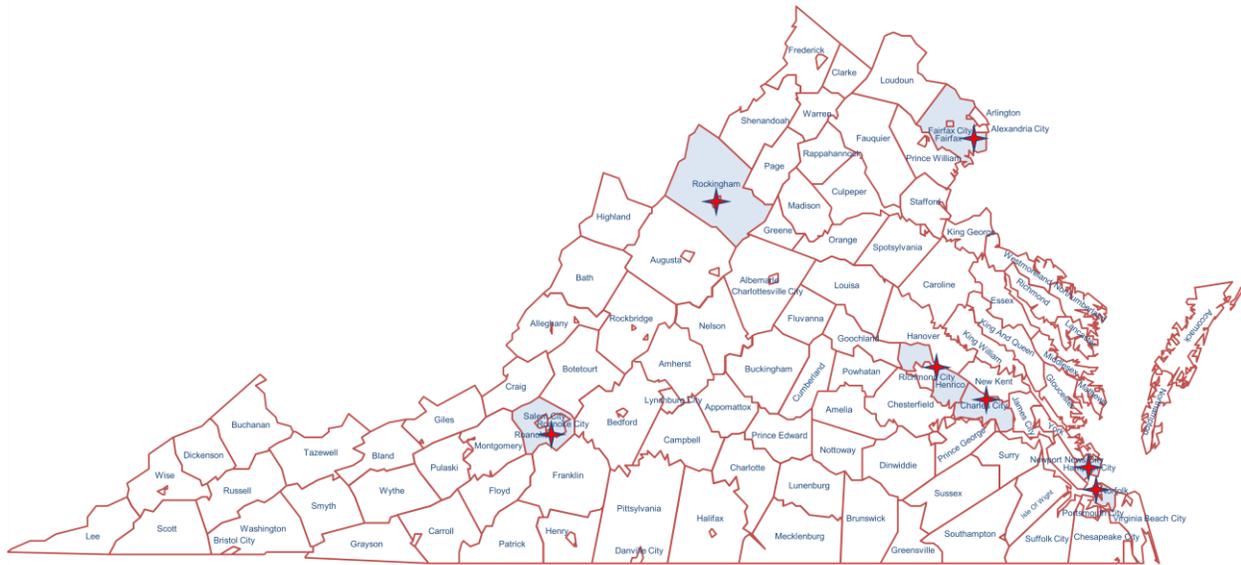


[Sulfur dioxide](#) (SO₂) is a colorless gas that has a strong odor. It results from burning of fuels containing sulfur (such as coal and oil), petroleum refining, and smelting (extracting metals from ore), and it also occurs naturally from volcanic eruptions. SO₂ can dissolve in water vapor to produce sulfuric acid, and it can also interact with other gases and particles in the air to produce sulfate aerosols that are capable of traveling long distances in the atmosphere.

EPA has developed primary and secondary air quality standards for SO₂. The primary standards are designed to protect people from the health effects of sulfur dioxide gas, which include respiratory problems for people with asthma and for those who are active outdoors. Long-term exposure to high concentrations of sulfur dioxide gas can cause respiratory illness and aggravate existing heart conditions. Sulfate particles that are formed from SO₂ gas can be inhaled, and are associated with increased respiratory symptoms and disease.

Secondary standards for sulfur dioxide protect against damage to vegetation and buildings, and against decreased visibility. The acids that can form from SO₂ and water vapor contribute to acid deposition (commonly called "acid rain") which causes damage to the leaves of plants and trees, making them vulnerable to disease, and can increase the acidity of lakes and streams, making them unsuitable for aquatic life. Acid deposition also causes deterioration of materials on buildings, monuments, and sculptures. Finally, small sulfate particles, formed when SO₂ gas reacts with other gases and particles in the air, contribute to haze that causes decreased visibility in many areas of the country.

Sulfur dioxide is monitored continuously with an electronic instrument using ultraviolet fluorescence detection. The instrument has a pump that pulls outside air into a sample chamber containing a high intensity ultraviolet (UV) light. Any SO₂ molecules in the sample air absorb some of the UV light, become excited, and then fluoresce, releasing light characteristic of SO₂. The fluorescence is detected with a photomultiplier tube (a tube that detects very small amounts of light and multiplies the signal many times), and the resulting signal, which corresponds to the amount of SO₂ in the sample, is converted to an SO₂ concentration by the instrument computer.



VA Department of Environmental Quality

SO2 Monitoring Network

National Ambient Air Quality Standards (NAAQS)

Primary Standards for SO₂:

- 3-year average of the 99th percentile 1-hour daily maximum values not to exceed 75 ppb.

Secondary Standard for SO₂:

- 3-Hour concentration not to exceed 0.5 ppm (500 ppb) more than once per year.

Sulfur Dioxide 99 th Percentile 1-Hour Daily Maximum Values (ppb)					
Site ID	City/County	2014	2015	2016	3-Yr Avg Design Value 2014-2016
26-F	Rockingham Co.	7	3	2	4
19-A6	Roanoke Co.	6	5	4	5
72-M	Henrico Co.	8	8	5	7
75-B	Charles City Co.	29*	29	24	27
179-K	Hampton	41	30	17*	29
181-A1	Norfolk	36	13	9	19
46-B9	Fairfax Co.	11	9	5	8

* Did not meet completeness criteria

National Ambient Air Quality Standards (NAAQS)Primary Standards for SO₂:

- 3-year average of the 99th percentile 1-hour daily maximum values not to exceed 75 ppb.

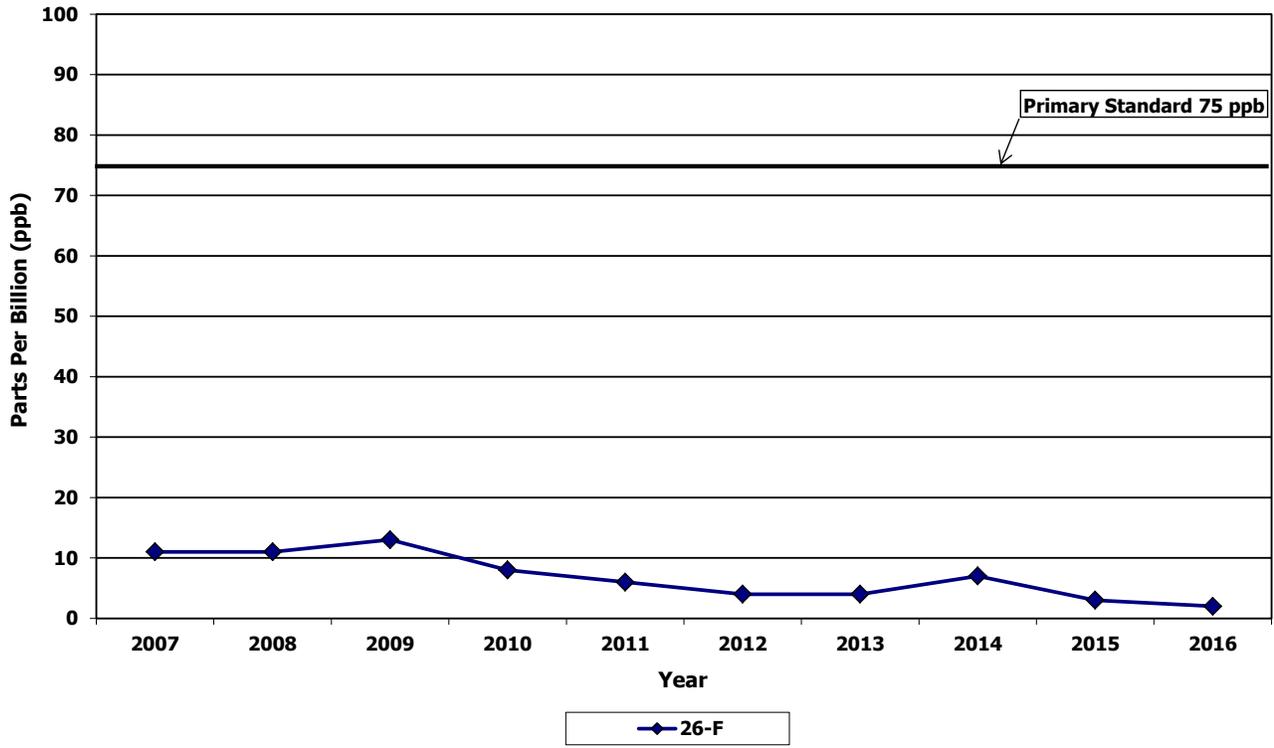
Secondary Standard for SO₂:

- 3-Hour concentration not to exceed 0.5 ppm (500 ppb) more than once per year.

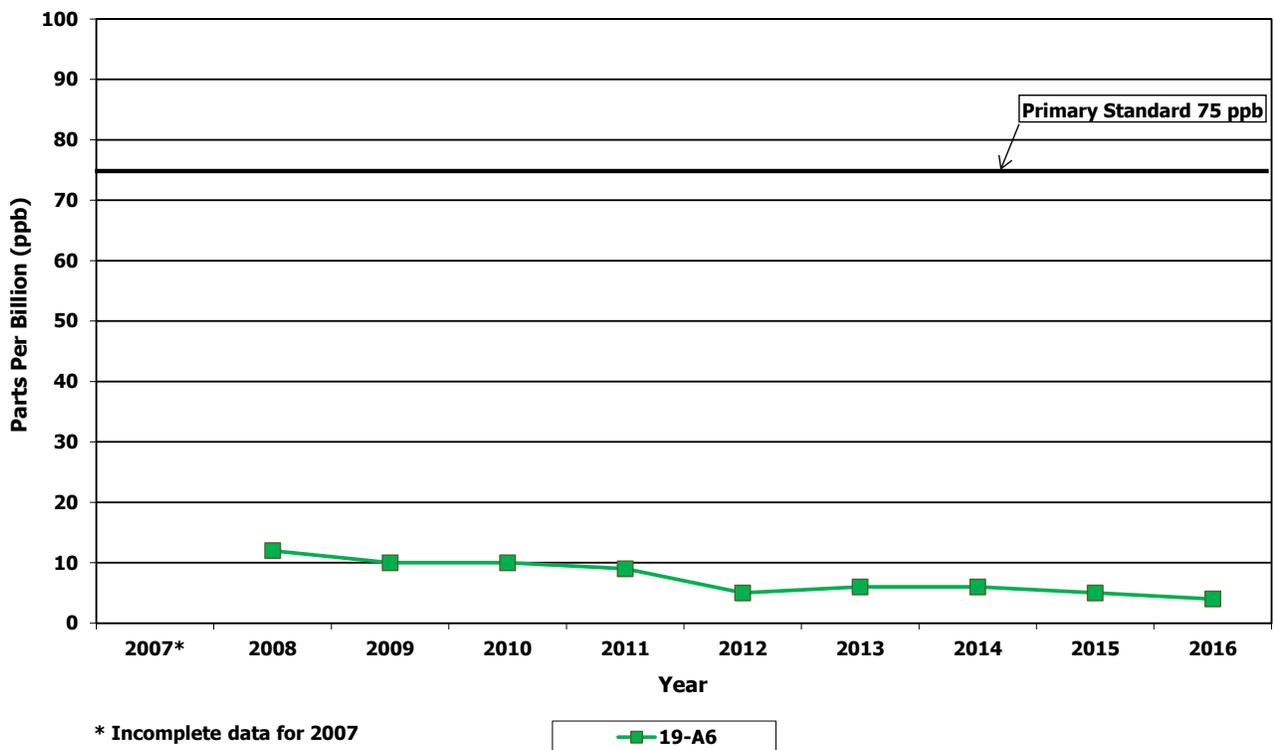
Sulfur Dioxide 3-Hour Block Average Maximum Values (ppb)					
Site ID	City/County	2014	2015	2016	Number Obs. > 500 ppb
26-F	Rockingham Co.	7	5	2	0
19-A6	Roanoke Co.	6	7	4	0
72-M	Henrico Co.	10	7	6	0
75-B	Charles City Co.	27	37	16	0
179-K	Hampton	29	28	15	0
181-A1	Norfolk	25	16	13	0
46-B9	Fairfax Co.	13	10	6	0

SO₂ Monitoring Network

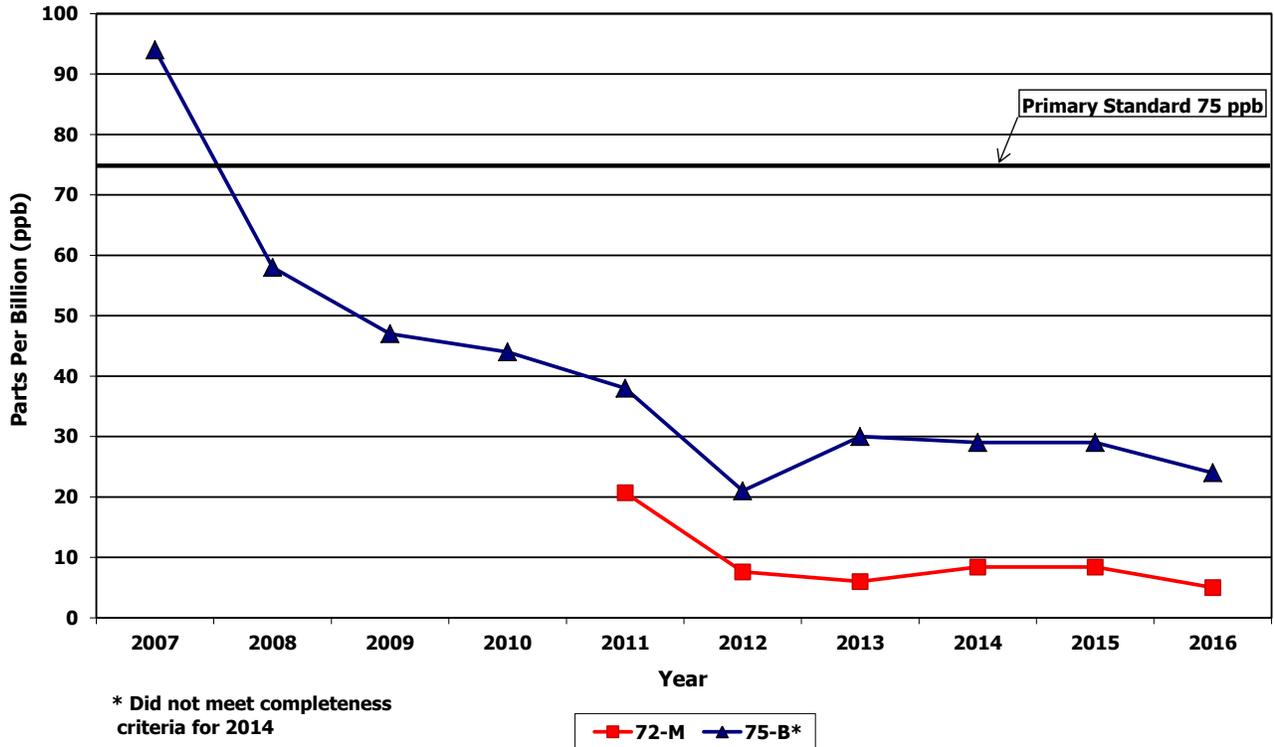
Sulfur Dioxide - Valley Region 99th Percentile 1-Hour Daily Maximum



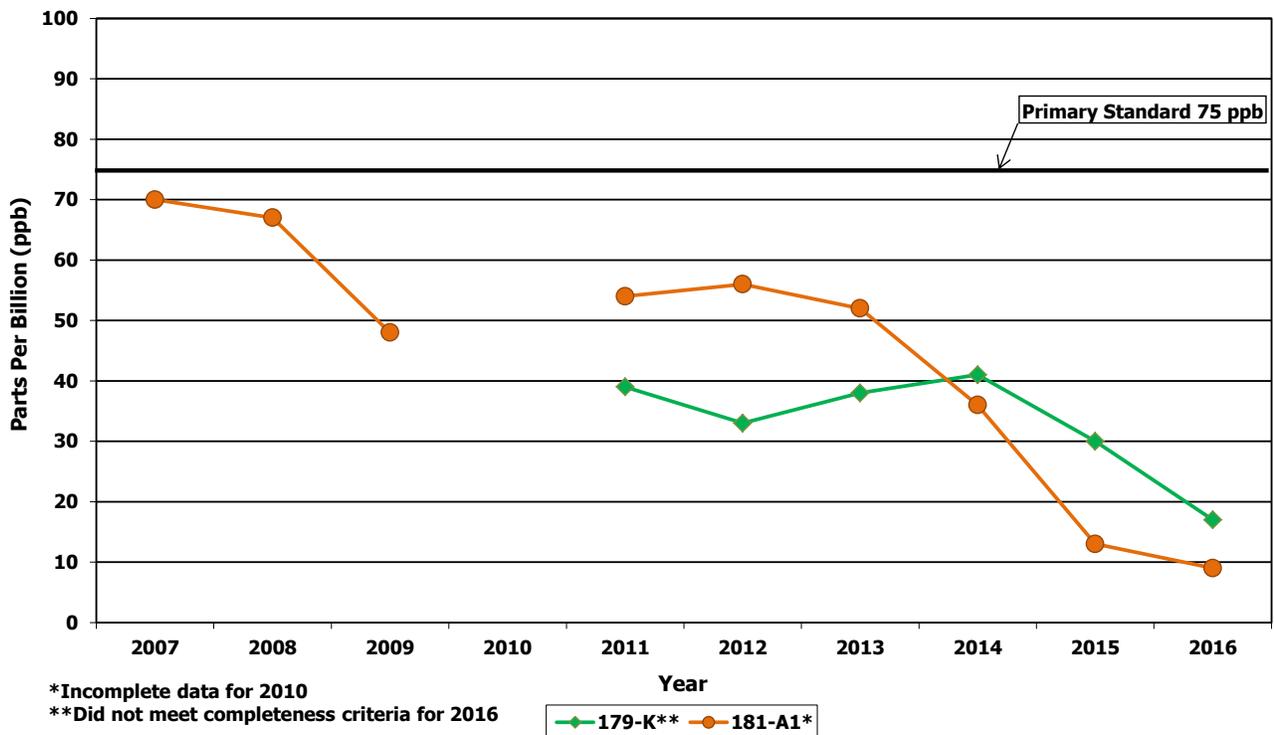
Sulfur Dioxide - Blue Ridge Region 99th Percentile 1-Hour Daily Maximum



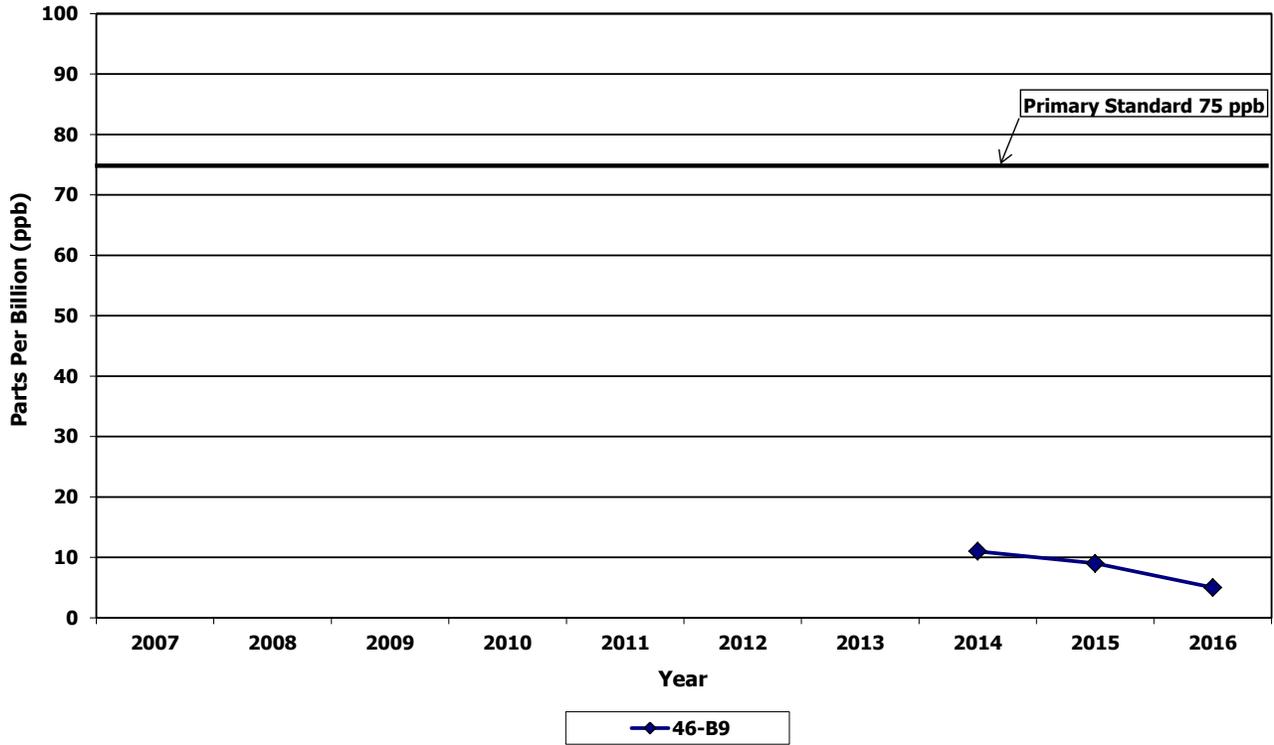
Sulfur Dioxide - Piedmont Region 99th Percentile 1-Hour Daily Maximum



Sulfur Dioxide - Tidewater Region 99th Percentile 1-Hour Daily Maximum



Sulfur Dioxide - Northern Region 99th Percentile 1-Hour Daily Maximum

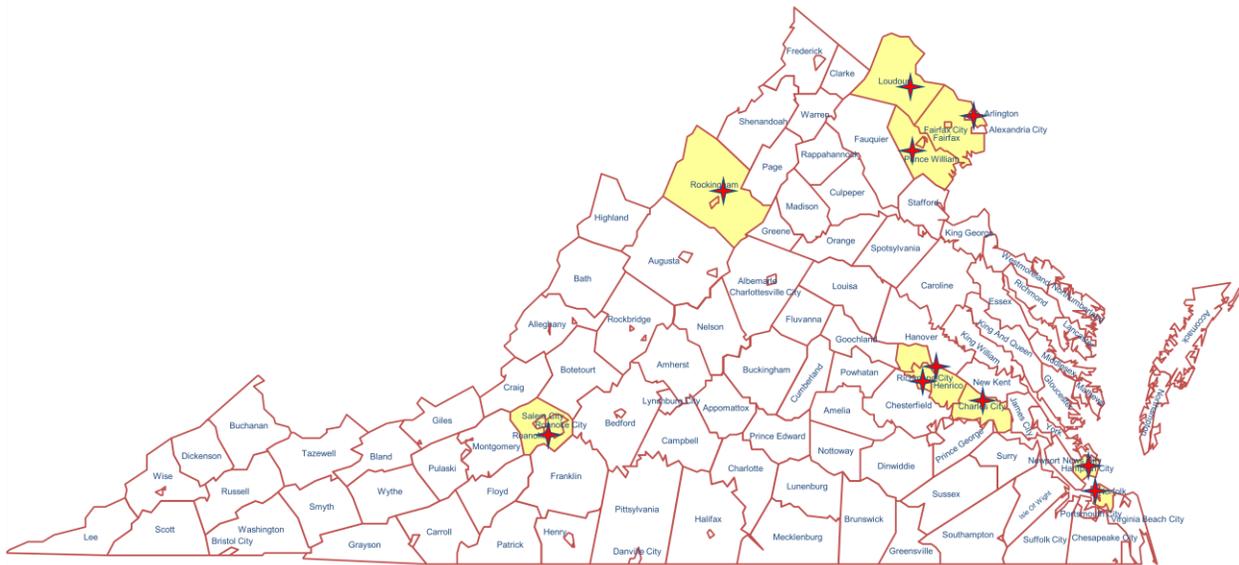


[Nitrogen dioxide](#) (NO₂) is one in a group of gases referred to as oxides of nitrogen (NO_x). Nitrogen dioxide, which is characterized by a reddish-brown color and pungent odor, along with the other NO_x gases, results from high-temperature burning of fossil fuels in automobiles, power plants, and other industrial, commercial, and residential sources. NO_x can occur naturally from lightning, forest fires, and bacterial processes that take place in soil.

NO_x pollution contributes to a wide range of problems in the environment. Ground-level ozone, a major component of “smog”, forms when NO_x and volatile organic compounds (VOCs) react in the presence of sunlight. NO_x also reacts with other gases and particles in the air to form acids that contribute to acid deposition, and to form small particles that can be inhaled into the lungs. NO_x contributes to water quality deterioration by depositing nitrogen into water bodies, upsetting the nutrient balance and causing oxygen depletion in the water so that fish and other aquatic life cannot survive. Nitrate particles and nitrogen dioxide also contribute to visibility impairment by blocking light transmission.

EPA has established primary and secondary air quality standards for NO₂ because it can cause lung irritation and respiratory problems in humans. Small particles formed from reaction of NO_x gases with other compounds can be inhaled deep into the lungs and cause or worsen respiratory conditions such as emphysema and bronchitis, and can aggravate existing heart conditions.

Nitrogen oxides are measured continuously with electronic instruments using the “gas phase chemiluminescence” method. The instrument has a pump that draws ambient air into a reaction chamber. Inside the chamber, the air is mixed with a high concentration of ozone (O₃). Any nitric oxide (NO) present in the sample air reacts with O₃ to produce NO₂. The NO₂ molecules created by the reaction are in an excited state, and emit light characteristic of NO₂ – this is called “chemiluminescence.” The light produced in the reaction is detected with a photomultiplier tube, and the resulting signal is converted to a number reflecting the concentration of NO in the ambient air by the instrument computer. The instrument then activates a valve that diverts incoming ambient air into a “converter”, which converts any NO₂ in the ambient air to NO by reduction reaction. After the air passes through the converter, it is sent to the reaction chamber where the NO and O₃ react to produce NO₂. The chemiluminescence produced by the reaction is converted to a signal that reflects the concentration of NO_x in the ambient air. The instrument then calculates the NO₂ concentration using the difference between the measured NO_x and NO concentrations.



VA Department of Environmental Quality

NO2 Monitoring Network

National Ambient Air Quality Standards (NAAQS)

Primary Standard for NO₂:

- 3-year average of the 98th percentile 1-hour daily maximum values not to exceed 100 ppb.
- Annual Arithmetic Mean not to exceed 53 ppb (.053 ppm).

Secondary Standard for NO₂:

- Same as primary.

Nitrogen Dioxide 98th Percentile 1-Hour Daily Maximum Values (ppb)				
Site City/County	2014	2015	2016	3-Yr Avg. Design Value 2014-2016
(26-F) Rockingham Co.	42	42	35	40
(19-A6) Roanoke Co.	39	31	37	36
(72-M) Henrico Co.	37*	35	36*	36
(75-B) Charles City Co.	44	38	44	42
(158-X) Richmond	45*	47	44	45
(179-K) Hampton	29	28	27	28
(181-A1) Norfolk	42	41	39	41
(38-I) Loudoun Co.	43*	45*	35	41
(45-L) Prince William Co.	31	26	27	28
(47-T) Arlington Co.	50*	49	44	48

* Did not meet completeness criteria

National Ambient Air Quality Standards (NAAQS)Primary Standard for NO₂:

- 3-year average of the 98th percentile 1-hour daily maximum values not to exceed 100 ppb.
- Annual Arithmetic Mean not to exceed 53 ppb (.053 ppm).

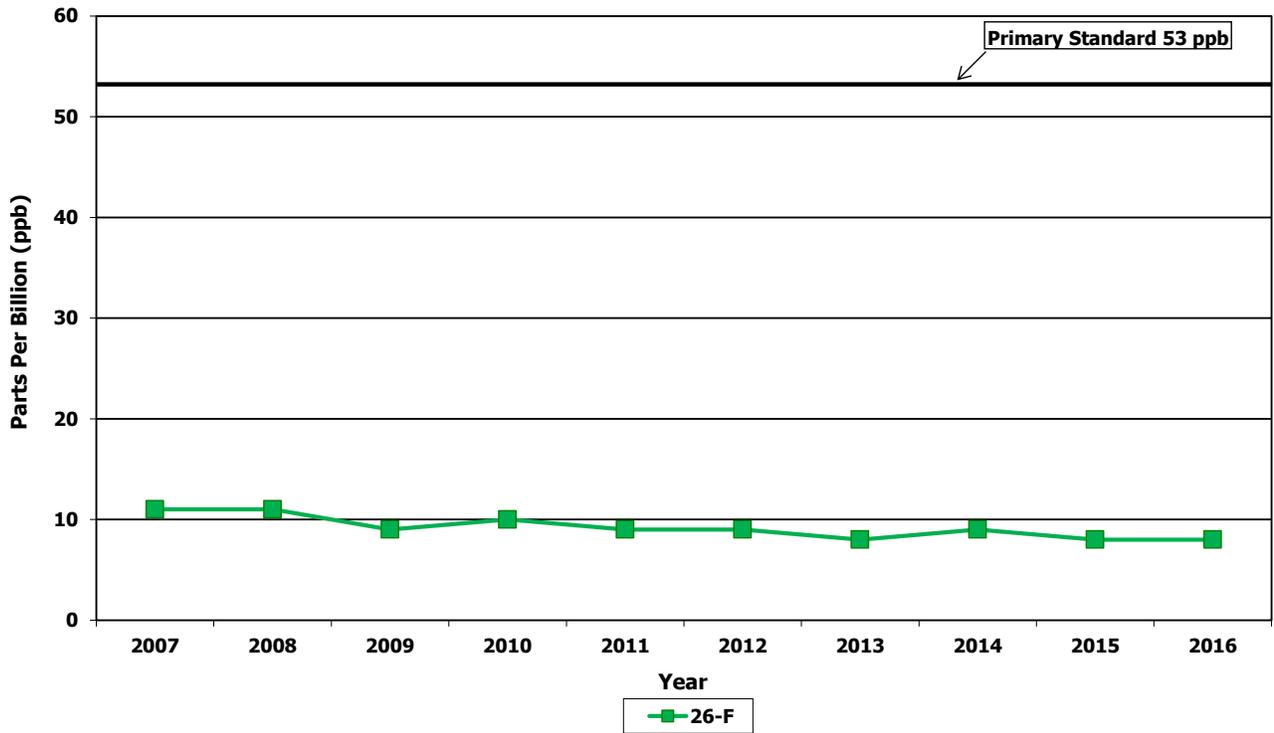
Secondary Standard for NO₂:

- Same as primary.

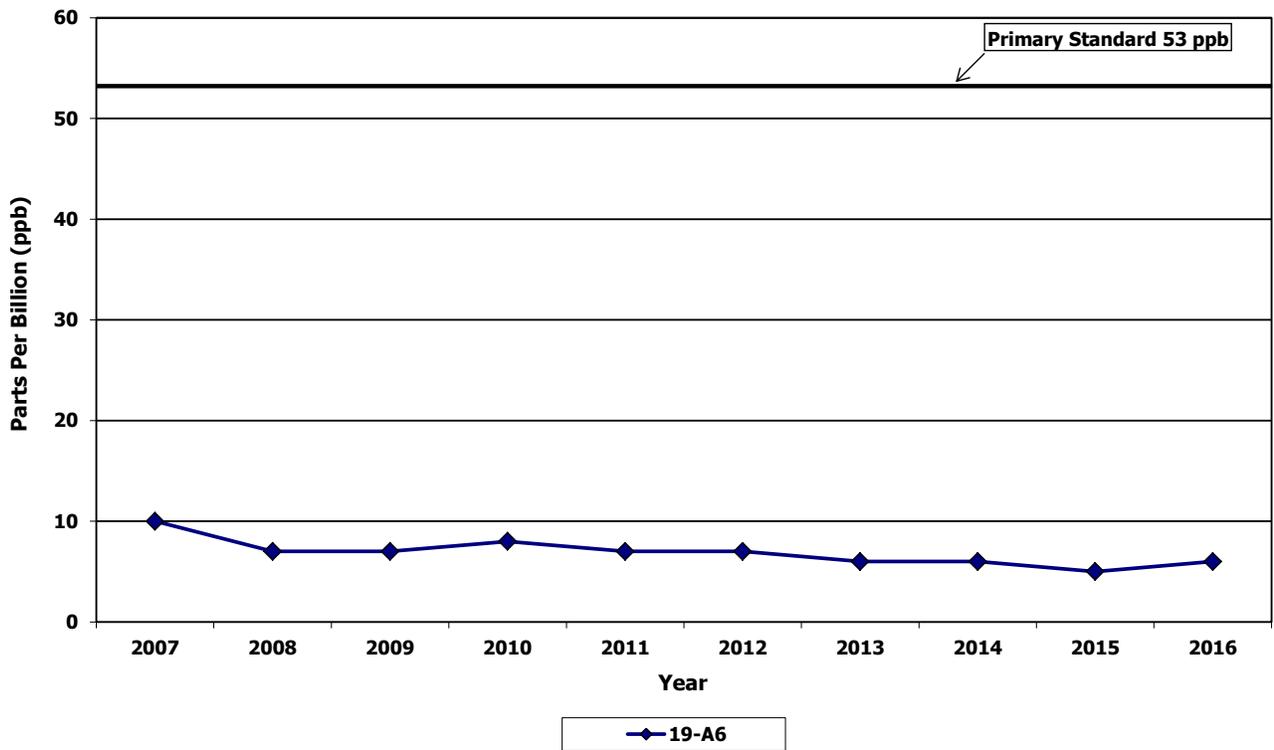
Site	Annual Arithmetic Mean (ppb)		
	2014	2015	2016
(26-F) Rockingham Co.	9	8	8
(19-A6) Roanoke Co.	6	5	6
(72-M) Henrico Co.	8	8	7
(75-B) Charles City Co.	5	4	3
(158-X) Richmond	14*	14	12
(179-K) Hampton	4	4	3
(181-A1) Norfolk	8	8	7
(38-I) Loudoun Co.	7	8	7
(45-L) Prince William Co.	5	5	5
(47-T) Arlington Co.	11	11	11

* Did not meet completeness criteria

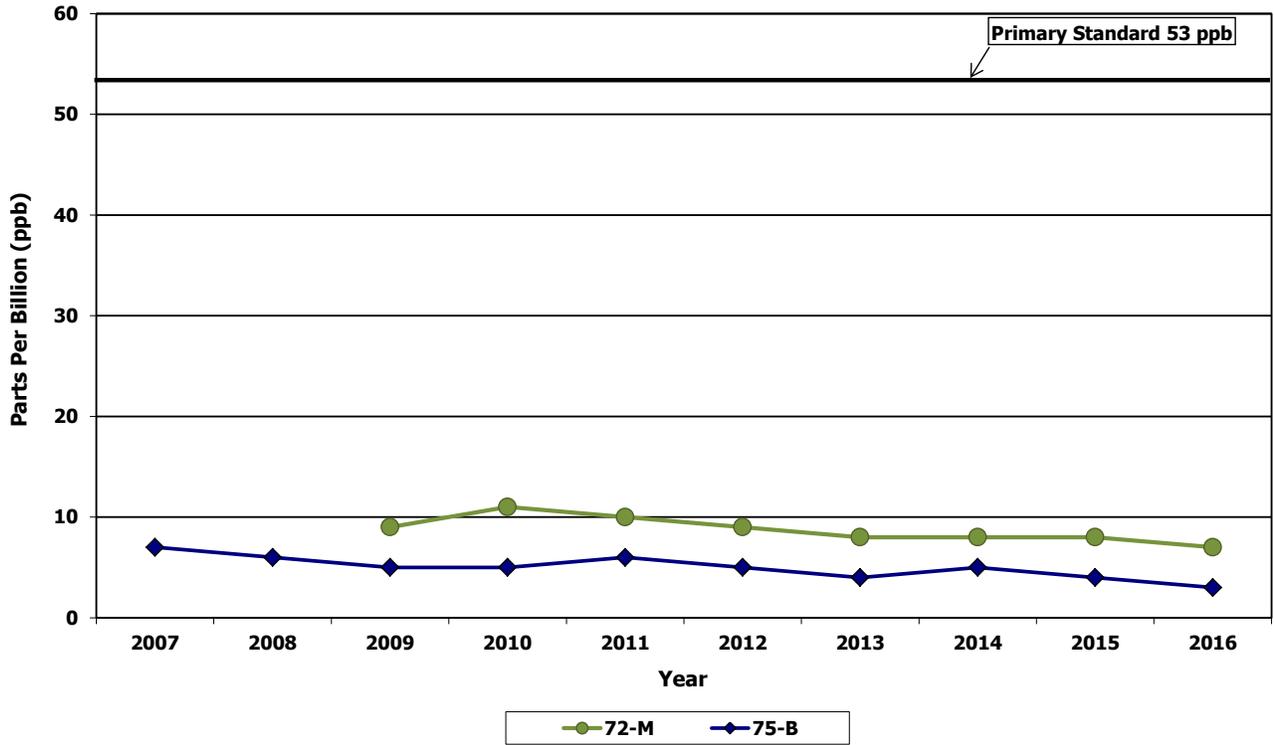
Nitrogen Dioxide - Valley Region Annual Arithmetic Mean



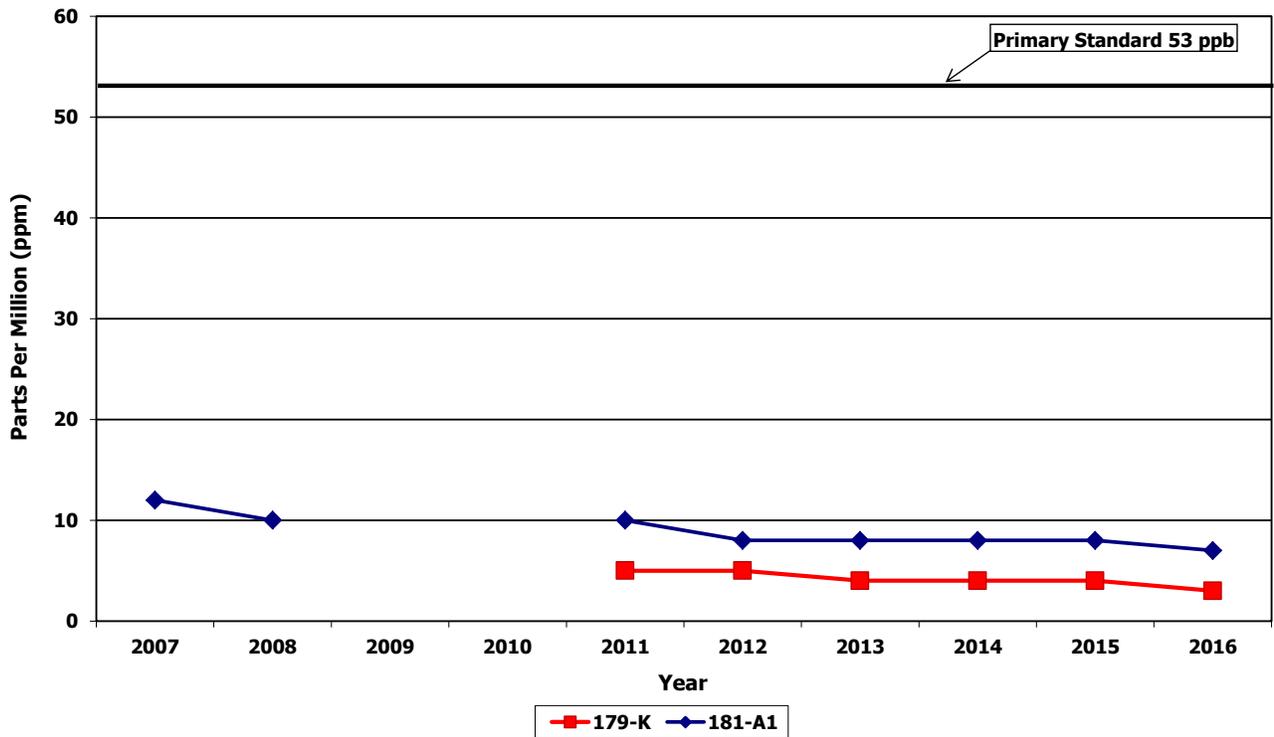
Nitrogen Dioxide - Blue Ridge Region Annual Arithmetic Mean



Nitrogen Dioxide - Piedmont Region Annual Arithmetic Mean

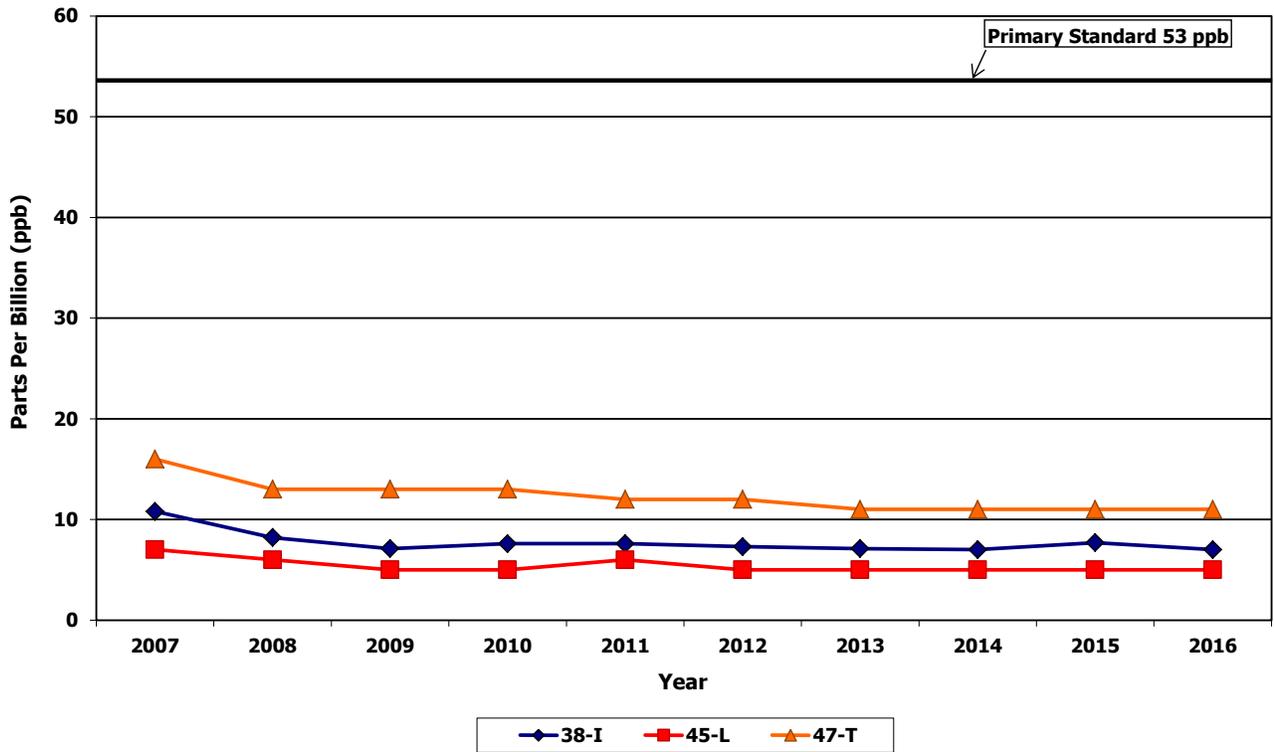


Nitrogen Dioxide - Tidewater Region Annual Arithmetic Mean



*Incomplete data in 2009 & 2010

Nitrogen Dioxide - Northern Region Annual Arithmetic Mean



Ozone (O_3) is a gas comprised of three oxygen atoms. It is unstable, and a strong oxidizing agent, and will react readily with other compounds to decay to the more stable diatomic oxygen (O_2).

Ozone is a gas that occurs both in the Earth's upper atmosphere and at ground level. Ozone can be "good" or "bad" for people's health and for the environment, depending on its location in the atmosphere. "Good" ozone occurs naturally in the stratosphere, about 10-30 miles above the earth's surface, where it forms a layer that filters the sun's ultraviolet rays before they reach the surface where they can cause harm to people, animals, and plants. "Bad" ozone, or ground-level ozone, occurs when chemicals found in the atmosphere at earth's surface react in the presence of intense sunlight. Ozone at ground level is harmful because it can cause a variety of health problems, as well as damage to plants and materials. Since ground-level ozone is not emitted directly, it is called a "secondary" pollutant. The chemicals needed to form ozone, NO_x and hydrocarbons (also called volatile organic compounds, or VOCs), can come from motor vehicle exhaust, power plants, industrial emissions, gasoline vapors, chemical solvents, as well as natural sources such as lightning, forest fires, and plant decomposition. Ozone, and the chemicals that produce ozone, can travel hundreds of miles from their sources, so that even rural areas with few pollutant emissions can occasionally experience high ozone levels. Efforts to control ground-level ozone involve limiting emissions of NO_x and VOCs, or "ozone precursors," that are necessary for ozone production.

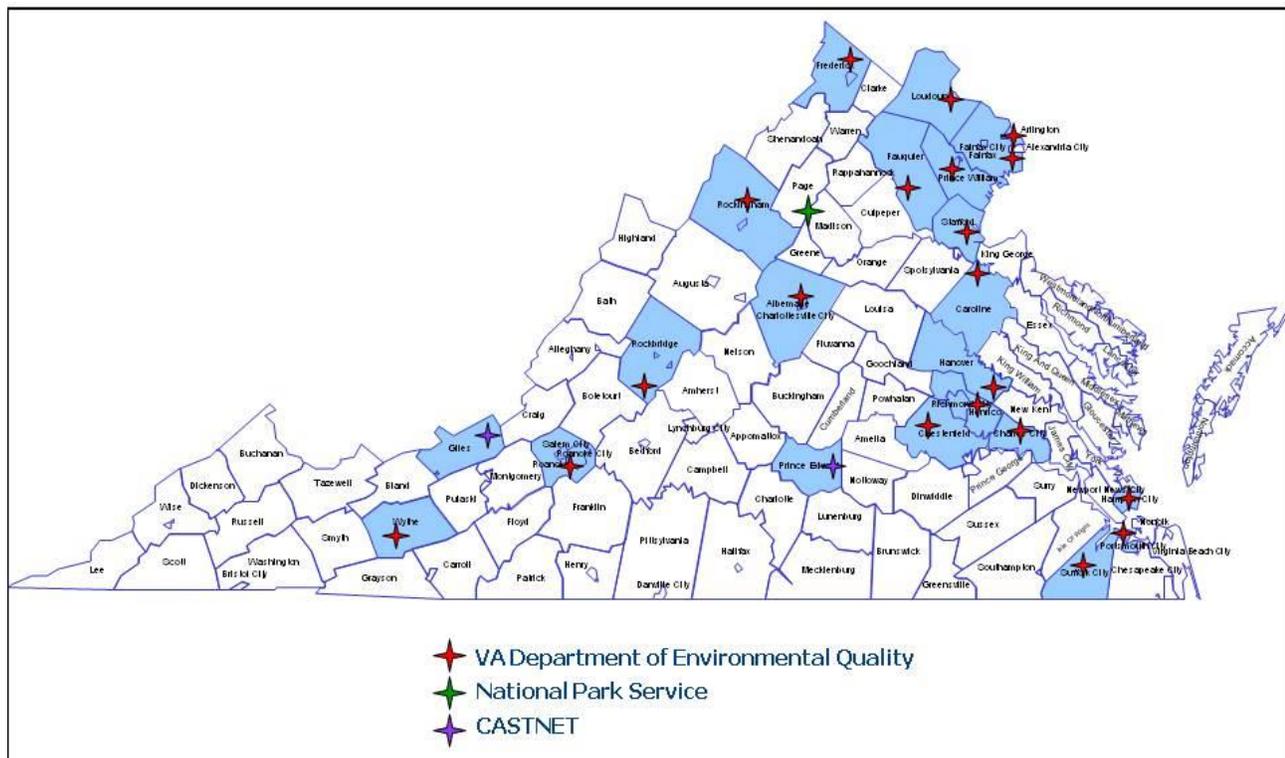
Ground-level ozone is a seasonal pollutant, and the length of the ozone season varies across the country. In some areas, the season may last most of the year, but in Virginia it is usually only a problem during the late spring to summer months when the sunlight is most intense. For 2016, Virginia was required to operate its ozone monitors from the months of April to October. In addition to the seasonal pattern, ozone also has a strong diurnal (daily) pattern at low altitudes, so that it is usually depressed at night, but begins to build during the day after the sun rises.

EPA has created primary and secondary air quality standards for ground-level ozone because of its adverse effects on public health and welfare. In humans, ozone can irritate lung airways, causing sunburn-like inflammation, and can induce symptoms such as wheezing, coughing, and pain when taking a deep breath. Although people with existing respiratory problems, such as asthma and emphysema, are most vulnerable, young children and otherwise healthy people can also suffer respiratory problems from ozone exposure. Scientific studies have shown that even at low levels, ozone can trigger breathing problems for sensitive individuals. In addition to human health problems, ozone can damage the leaves of plants and trees, making them susceptible to disease, insects, and harsh weather. Ozone can also cause rubber to harden and crack, and some painted surfaces to fade more quickly.

Ozone is measured continuously with electronic instruments using "ultraviolet (UV) absorption photometry." The method is based on the principle that ozone strongly absorbs UV light at 254 nanometers (a nanometer is equal to a distance of one billionth of a meter). The ozone monitor has a sample pump that draws ambient air into it and splits the air into two gas streams. In one stream, the air passes through an "ozone scrubber", which cleanses the sample air of any ozone. Then the clean air passes through a sample cell that contains a UV light source and a detector. The detector measures the intensity of the light in the sample cell, providing a zero reference. The second air stream is sent straight into the sample cell, bypassing the scrubber. Any ozone present in the incoming air will absorb some of the UV light in the sample cell, reducing the amount of light reaching the detector. The instrument then calculates the ozone concentration of the ambient air from the difference in the light intensities measured between the scrubbed, or "zero" air, and the unscrubbed air.

Daily ozone forecasts for selected metropolitan areas and hourly ozone values for all Virginia ozone monitoring sites can be viewed for the months of April to October on the DEQ web page at http://vadeq.tx.sutron.com/cgi-bin/aqi_rpt.pl. In addition, animated ozone maps for Virginia and other parts of the United States are available at <http://www.airnow.gov/>.

The National Park Service operated one ozone monitor at Big Meadows in Shenandoah National Park in 2015. Daily data from this site are available at the DEQ website, and historical data may be obtained on the internet at <http://ard-request.air-resource.com> or at EPA's AirData website. EPA also maintains Ozone monitoring sites in Giles and Prince Edward counties as part of the CASTNET program. Data from these sites can be viewed at <https://www.epa.gov/castnet>.



National Ambient Air Quality Standards (NAAQS)

Primary Standard for O₃:

- Maximum 8-hour average concentration of 0.070 ppm, effective October 1, 2015, based on 3-year average of the annual fourth highest daily maximum 8-hour averages.

- Secondary Standard for O₃:

Same as primary

The standard is attained at a monitoring site when the 3-year average of the fourth highest daily maximum 8-hour average for each of the three most recent years is less than or equal to 0.070 ppm.

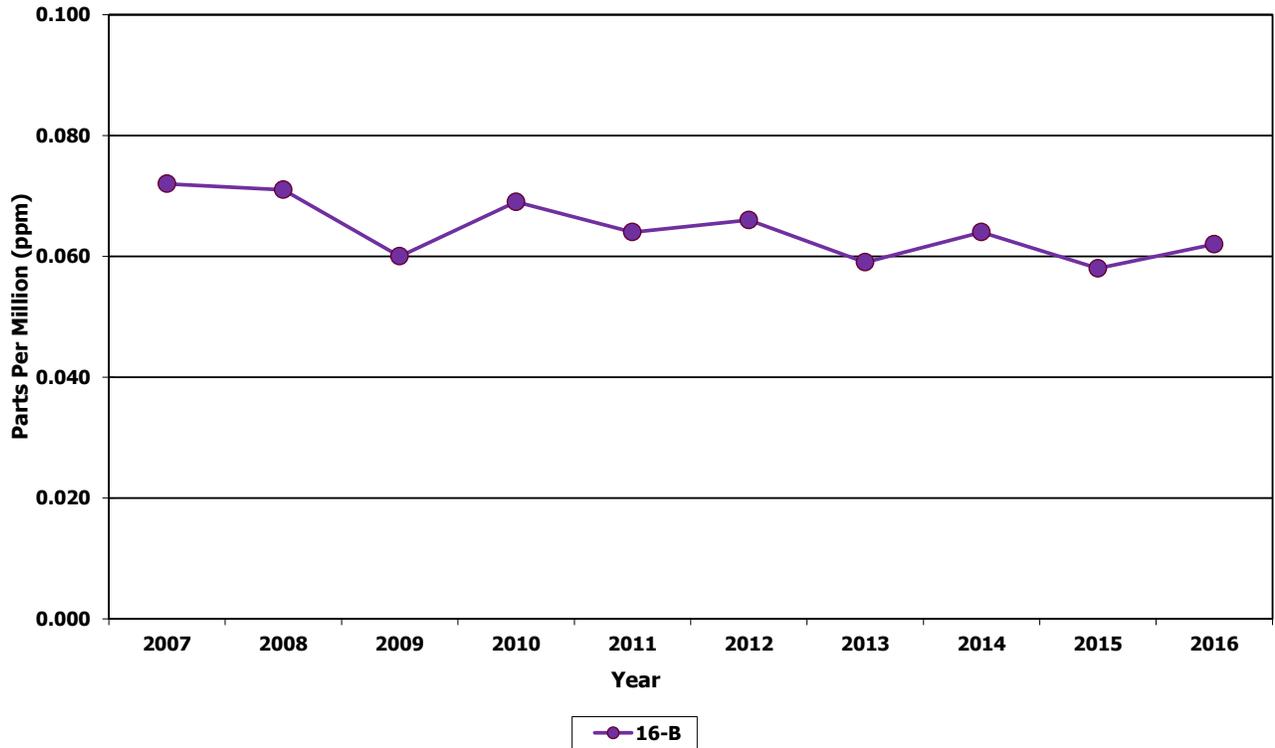
Site	Days Exceeded 0.070 ppm	2016			
		Highest Daily Maximum 8-Hour Avg.			
		1 st Max.	2 nd Max.	3 rd Max.	4 th Max.
(16-B) Wythe Co.	0	.070	.065	.064	.062
(26-F) Rockingham Co.	1	.073	.069	.062	.062
(28-J) Frederick Co.	0	.067	.066	.065	.065
(33-A) Albemarle Co.	1	.077	.070	.068	.062
(19-A6) Roanoke Co.	0	.070	.067	.065	.064
(21-C) Rockbridge Co.	0	.067	.064	.061	.060
(71-H) Chesterfield Co.	1	.075	.067	.065	.063
(72-M) Henrico Co.	3	.079	.076	.072	.066
(73-E) Hanover Co.	2	.072	.072	.066	.065
(75-B) Charles City Co.	1	.074	.067	.067	.065
(179-K) Hampton	2	.080	.074	.069	.068
(183-E) Suffolk	1	.073	.070	.064	.063
(183-F) Suffolk	1	.075	.065	.064	.061
(37-B) Fauquier Co.	0	.068	.067	.064	.063
(38-I) Loudoun Co.	1	.074	.069	.069	.068
(44-A) Stafford Co.	1	.073	.067	.066	.066
(45-L) Prince William Co.	3	.074	.072	.072	.067
(46-B9) Fairfax Co.	5	.079	.079	.073	.073
(47-T) Arlington Co.	6	.080	.076	.074	.072
(48-A) Caroline Co.	1	.074	.067	.064	.062

2014-2016 Fourth-Highest Daily Maximum 8-Hour Ozone Averages (units parts per million)					
	Monitor Location (County/City)	2014	2015	2016	3-Year Average (NAAQS = .070 ppm)
Richmond Maintenance Area	Chesterfield Co.	.061	.063	.063	.062
	Henrico Co.	.062*	.064	.066	.064
	Hanover Co.	.062	.061	.065	.062
	Charles City Co.	.066	.059	.065	.063
Hampton Roads Maintenance Area	Hampton City	.061	.065	.068	.064
	Suffolk City (TCC)	.058	.061	.063	.060
	Suffolk City (Holland)	.063	.060	.061	.061
Fredericksburg Maintenance Area	Stafford Co.	.062	.063	.066	.063
Northern Virginia Nonattainment Area	Loudoun Co.	.063	.071	.068	.067
	Prince William Co.	.062	.067	.067	.065
	Arlington Co.	.071	.073	.072	.072
	Fairfax Co. (Lee Park)	.065	.072	.073	.070
Shenandoah National Park Maintenance Area	Madison Co. (Big Meadows)	.060	.063	.066	.063
Areas Currently Designated Attainment	Wythe Co.	.064	.058	.062	.061
	Rockbridge Co.	.058	.056	.060	.058
	Rockingham Co.	.058	.060	.062	.060
	Frederick Co.	.059	.061	.065	.061
	Albemarle Co.	.059*	.059	.062	.060
	Roanoke Co.	.060	.062	.064	.062
	Fauquier Co.	.059	.056	.063	.059
	Caroline Co.	.061	.062	.062	.061

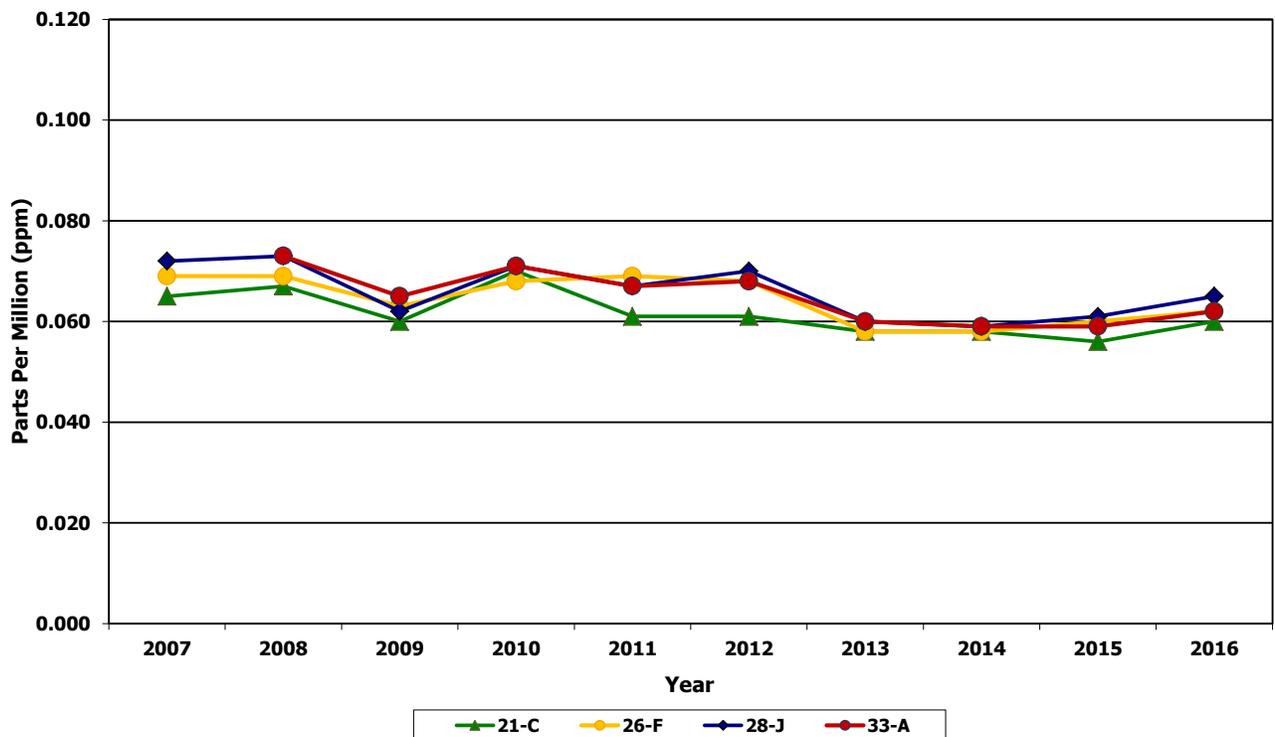
* Did not meet completeness criteria

A 3-year average of .070 ppm or above exceeds the 8-hour NAAQS for ozone.

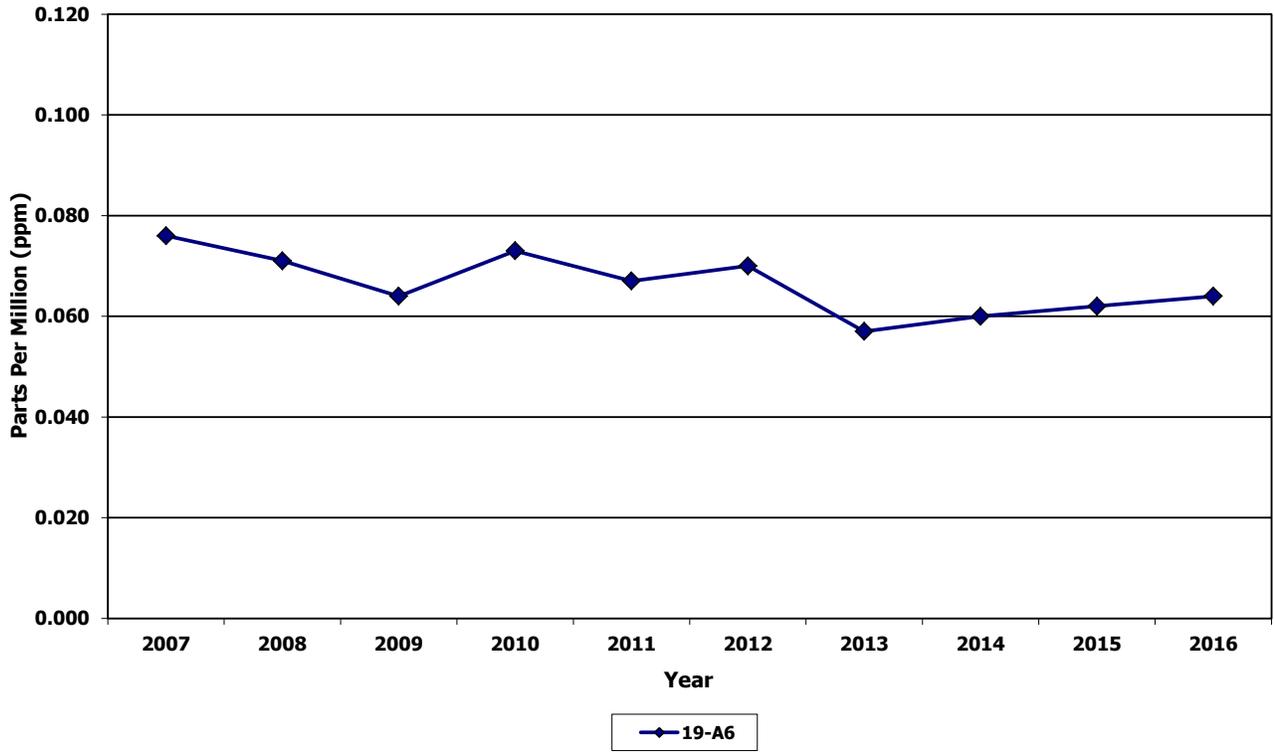
Ozone - Southwest Region 4th Daily Maximum, 8-Hour Value



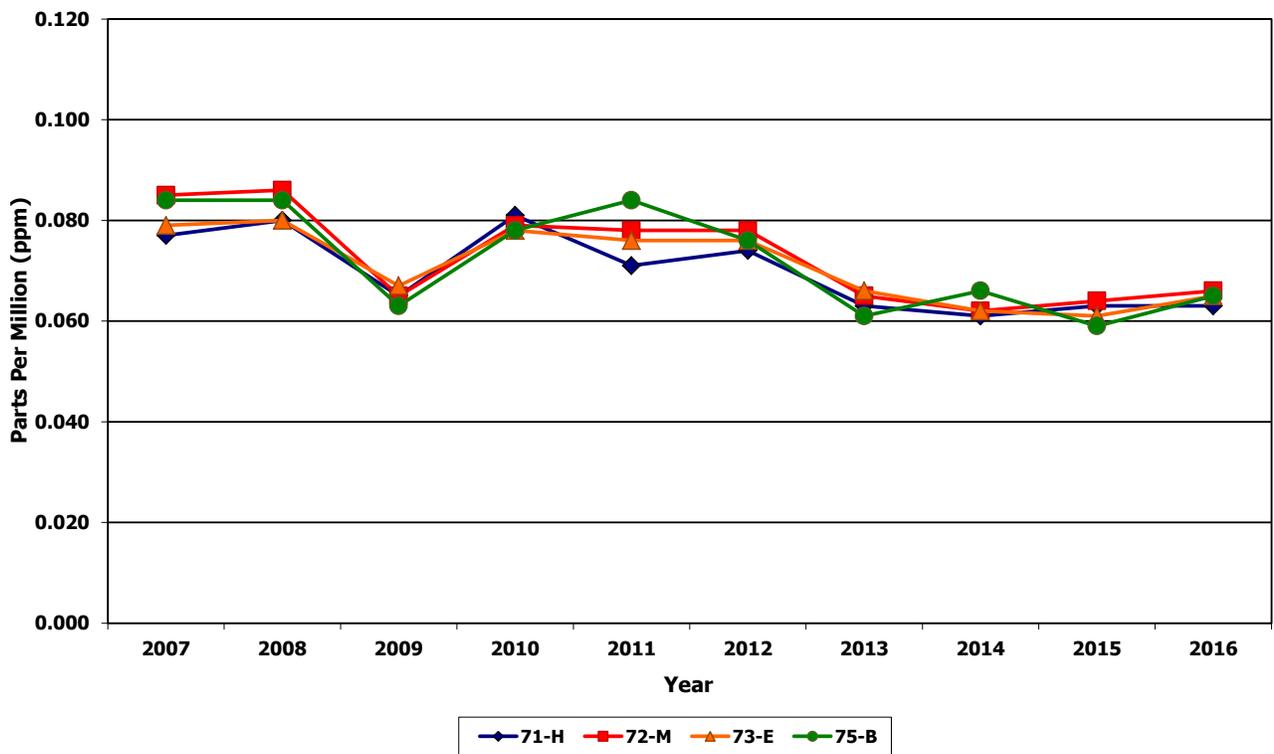
Ozone - Valley Region 4th Daily Maximum, 8-Hour Value



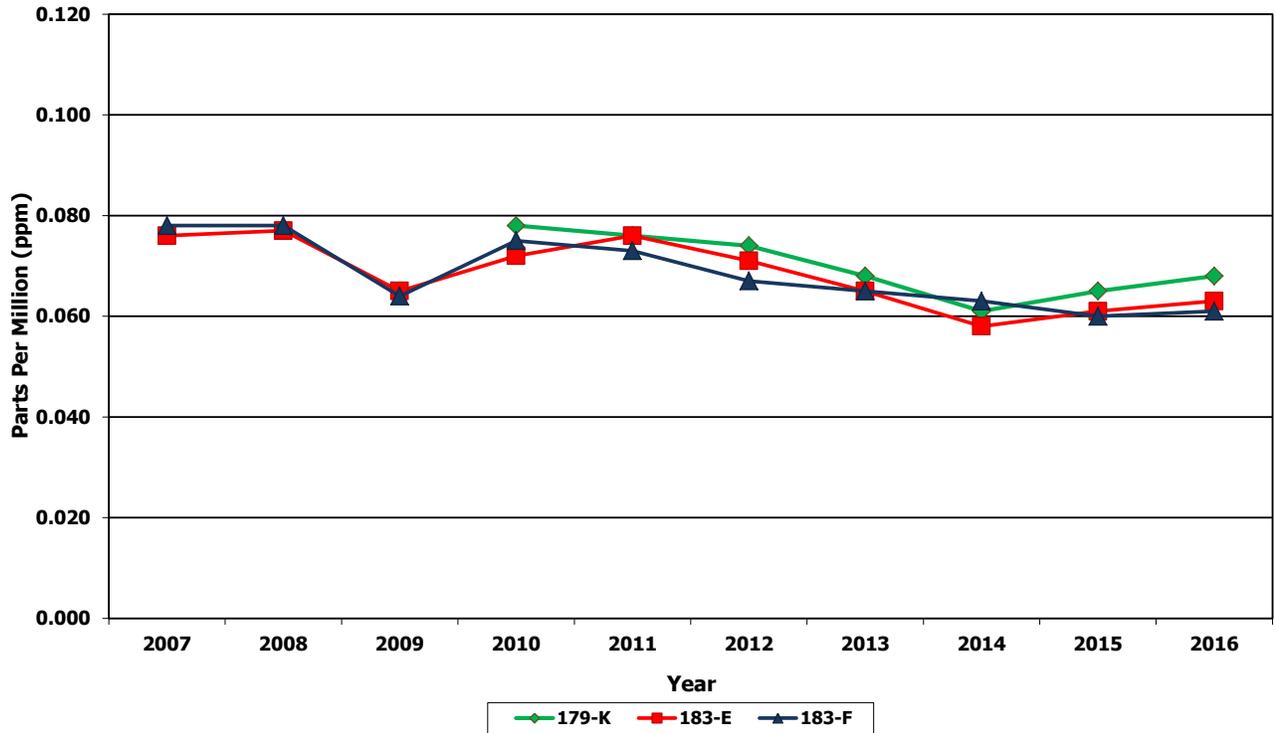
Ozone - Blue Ridge Region 4th Daily Maximum, 8-Hour Value



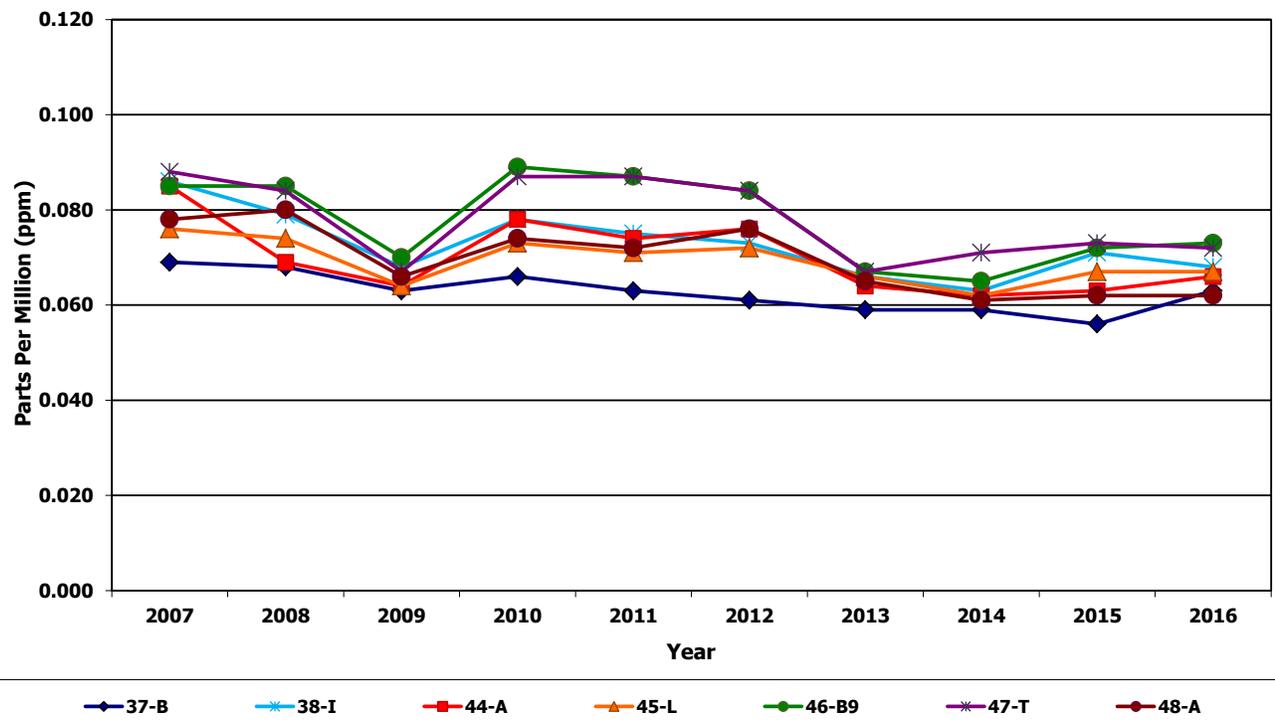
Ozone - Piedmont Region 4th Daily Maximum, 8-Hour Value



Ozone - Tidewater Region 4th Daily Maximum, 8-Hour Value



Ozone - Northern Region 4th Daily Maximum, 8-Hour Value



Lead (Pb) is a metal found naturally in the environment as well as in manufactured products. In the past, emissions from cars and trucks using leaded gasoline were the primary sources of lead in the atmosphere. Efforts by EPA to remove lead from motor vehicle gasoline resulted in dramatic reductions of lead in the ambient air from 1980 to 1999. Now the major sources of lead in the air are ore and metals processing and piston-engine aircraft operating on leaded aviation gasoline.

Particles containing lead can be inhaled, or lead can be ingested from drinking water or through contaminated food as a result of deposition of leaded particles onto the ground or in the water. In the body, lead can accumulate in the bones; affect the nervous system, kidney function, immune system, reproductive and developmental systems and the cardiovascular system. Young children are particularly vulnerable to the effects of lead, where it can contribute to behavioral problems, learning deficits and lowered IQ. Lead can stay in the environment for a long time, causing adverse effects to plants and animals.

The National Ambient Air Quality Standards, or NAAQS, for lead were revised in October 2008. At that time, EPA reduced the level of the standard from 1.5 micrograms per cubic meter to 0.15 micrograms per cubic meter. The secondary standard was also reduced to the level of the new primary standard. Virginia DEQ received a waiver from EPA in 1997 to discontinue lead monitoring because Virginia had no major lead sources. However, when the new standards were promulgated, the emission threshold that agencies were required to use for determining if a lead monitor was needed near a source also changed. As a result, Virginia had to resume monitoring for lead in a few areas, and AQM began installing the lead monitors in late 2009 and completed installation in October 2010. For additional information on the revised lead standards, see <https://www.epa.gov/lead-air-pollution/national-ambient-air-quality-standards-naaqs-lead-pb> .

To measure lead, ambient air is drawn into a high volume sampler. The sample air flows across an 8 x 10 inch glass fiber filter at a rate of 39-60 cubic feet per minute for a 24-hour period. The filter is sent to the Division of Consolidated Laboratories, where a small portion of it is analyzed using inductively coupled plasma – mass spectrometry (ICP-MS). The resulting lead concentration is reported as micrograms per cubic meter ($\mu\text{g}/\text{m}^3$). The normal sampling schedule is once every sixth day from midnight to midnight. The lead sampling schedule for 2016 can be found at <http://www3.epa.gov/ttn/amtic/calendar.html>.



Lead Monitoring Sites

National Ambient Air Quality Standards (NAAQS)

Primary Standard for Pb:

- 0.15 $\mu\text{g}/\text{m}^3$ three-month rolling average

Secondary Standard for Pb:

- Same as Primary

2016 Pb 3-Month Averages (units in $\mu\text{g}/\text{m}^3$, LC)				
Site	No. 24-Hour Observations	1 st Max	2 nd Max	>0.15 $\mu\text{g}/\text{m}^3$
(53-G) Amherst Co.	60	0.03	0.03	0
(109-N) Roanoke	60	0.01	0.01	0

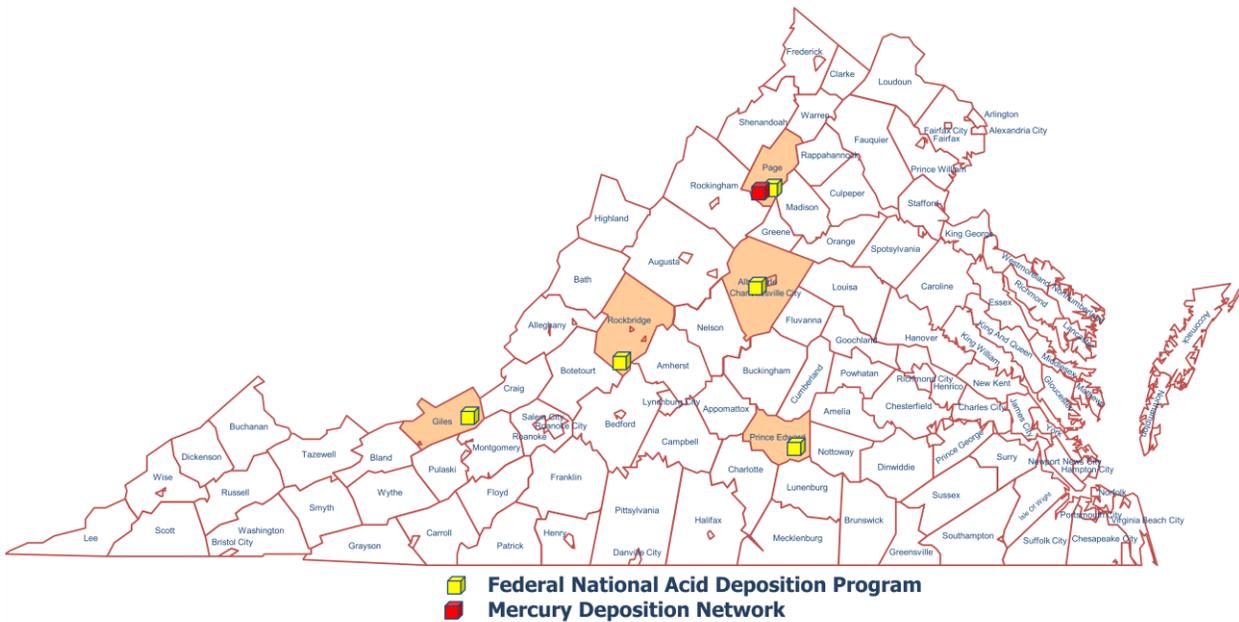
Acid Deposition Program

**Photochemical Assessment
Monitoring Stations**

Air Toxics Monitoring Network

The National Acid Deposition Program (NADP) had five monitoring sites in Virginia in 2016: Big Meadows (Shenandoah National Park), Hortons Station (Giles County), Charlottesville, Prince Edward County, and Natural Bridge Station (Rockbridge County). NADP site information and data are available on-line at <http://nadp.sws.uiuc.edu/ntn> in the NTN (National Trends Network) section.

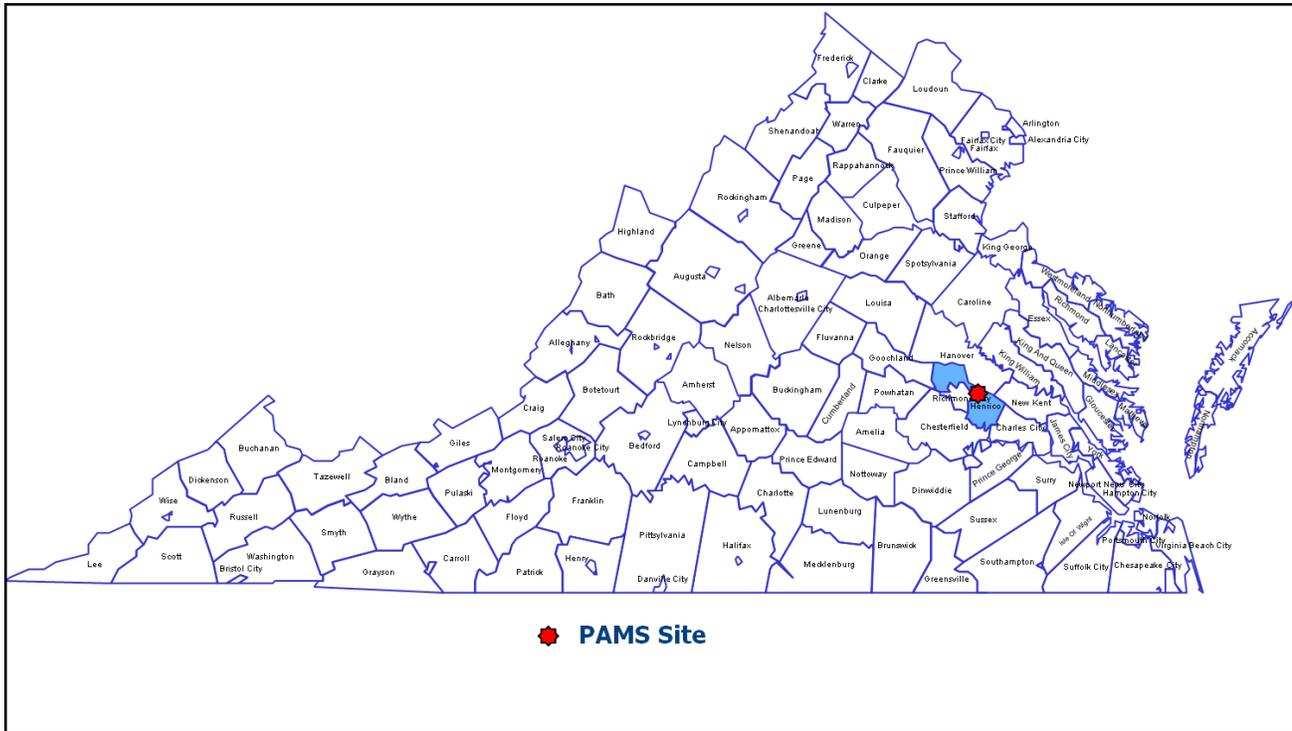
In addition to the five acid deposition monitors, there was one NADP Mercury Deposition Network (MDN) site in Virginia: Big Meadows (Shenandoah National Park). MDN site information and data are available on-line at <http://nadp.sws.uiuc.edu/MDN/>.



In 2016, the Office of Air Quality Monitoring (AQM) program of the Department of Environmental Quality collected 24-hour Photochemical Assessment Monitoring Station (PAMS) Volatile Organic Compounds (VOC) samples from MathScience Innovation Center (MSIC) in Henrico. Samples were collected using a one-in-six day sampling schedule.

AQM used the manual method for collecting ambient air samples. This method involves the collection of integrated, whole samples by using evacuated Summa^T or Silco^T canisters and Xonteck, Inc. air samplers. Each VOC sample from MSIC was analyzed by the Division of Consolidated Laboratory Services (DCLS) using a Gas Chromatograph/Flame Ionization Detector (FID) designated as TO-12. All VOC samples were analyzed for the presence of fifty-six target volatile organic precursors, and the measured concentration of Total Nonmethane Organic Compounds (TNMOC).

Detailed PAMS data are available upon request to the Virginia Department of Environmental Quality, Office of Air Quality Monitoring.



Photochemical Assessment Monitoring Network

**2016 Average Concentration of Detectable Volatile Ozone Precursors
Photochemical Assessment Monitoring Station (PAMS) Type II –
MathScience Innovation Center**

Concentrations are in ppbC
(non detects are counted as zeros for statistical purposes)

Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
43141	n-dodecane	53	0.00	0.71	0.20	0.281	0.219
43202	Ethane	53	0.00	33.83	5.82	9.046	8.144
43203	Ethylene	53	0.45	5.64	1.33	1.637	1.139
43204	Propane	53	1.28	13.84	3.75	4.992	3.081
43205	Propylene	53	0.17	2.26	0.53	0.667	0.451
43206	Acetylene	53	0.09	13.44	1.36	1.916	2.197
43212	n-butane	53	0.73	17.64	3.28	4.455	4.056
43214	Isobutane	53	0.00	3.77	0.65	0.874	0.847
43216	t-2-butene	53	0.00	0.61	0.00	0.050	0.119
43217	c-2-butene	53	0.00	2.68	0.04	0.193	0.507
43220	n-pentane	53	0.00	28.78	2.21	4.623	6.015
43221	Isopentane	53	0.34	13.61	2.86	4.193	3.204
43224	1-pentene	53	0.00	5.11	0.73	1.098	1.214
43226	t-2-pentene	53	0.00	7.84	0.24	0.761	1.346
43227	c-2-pentene	53	0.00	2.29	0.23	0.443	0.564
43230	3-methylpentane	53	0.13	2.59	0.53	0.709	0.560
43231	n-hexane	53	0.09	1.98	0.61	0.700	0.433
43232	n-heptane	53	0.08	0.98	0.35	0.395	0.223
43233	n-octane	53	0.00	0.70	0.18	0.213	0.145
43235	n-nonane	53	0.00	0.46	0.14	0.156	0.101
43238	n-decane	53	0.00	0.71	0.18	0.194	0.141
43242	Cyclopentane	53	0.14	1.47	0.35	0.411	0.247
43243	Isoprene	53	0.00	12.39	0.90	2.742	3.374
43244	2,2-dimethylbutane	53	0.00	0.55	0.14	0.161	0.105
43245	1-Hexene	53	0.00	0.71	0.07	0.153	0.188
43247	2,4-dimethylpentane	53	0.00	0.74	0.18	0.193	0.151
43248	Cyclohexane	53	0.00	0.86	0.23	0.258	0.216
43249	3-methylhexane	53	0.05	1.85	0.39	0.521	0.394
43250	2,2,4-trimethylpentane	53	0.16	3.43	0.77	0.992	0.752
43252	2,3,4-trimethylpentane	53	0.00	1.10	0.24	0.328	0.260
43253	3-methylheptane	53	0.00	0.47	0.13	0.133	0.093
43261	Methylcyclohexane	53	0.00	1.32	0.20	0.216	0.201
43262	Methylcyclopentane	53	0.14	1.89	0.43	0.515	0.328
43263	2-methylhexane	53	0.12	1.50	0.37	0.459	0.304
43280	1-butene	53	0.18	1.39	0.35	0.432	0.251
43284	2,3-dimethylbutane	53	0.00	1.56	0.29	0.403	0.343
43285	2-methylpentane	53	0.11	3.99	0.80	0.964	0.827
43291	2,3-dimethylpentane	53	0.00	0.74	0.18	0.227	0.166
43954	n-undecane	53	0.00	0.65	0.20	0.214	0.139
43960	2-methylheptane	53	0.00	0.41	0.09	0.102	0.079
45109	m/p-xylene	53	0.00	1.78	0.59	0.735	0.442
45201	Benzene	53	0.00	3.03	0.68	0.914	0.690
45202	Toluene	53	0.66	7.14	1.68	2.105	1.342
45203	Ethylbenzene	53	0.10	1.87	0.48	0.662	0.420
45204	o-xylene	53	0.00	1.17	0.27	0.346	0.245
45207	1,3,5-trimethylbenzene	53	0.00	0.62	0.13	0.161	0.138
45208	1,2,4-trimethylbenzene	53	0.00	1.48	0.41	0.517	0.319
45209	n-propylbenzene	53	0.00	0.64	0.12	0.148	0.139
45210	Isopropylbenzene	53	0.00	0.25	0.05	0.062	0.069
45211	o-ethyltoluene	53	0.00	1.27	0.16	0.235	0.258
45212	m-ethyltoluene	53	0.00	1.20	0.37	0.400	0.261
45213	p-ethyltoluene	53	0.00	1.28	0.21	0.260	0.252
45218	m-diethylbenzene	53	0.00	0.71	0.00	0.055	0.136
45219	p-diethylbenzene	53	0.00	0.85	0.00	0.079	0.156
45220	Styrene	53	0.00	3.28	0.47	0.820	0.832
45225	1,2,3-trimethylbenzene	53	0.00	0.77	0.07	0.120	0.165
43000	PAMHC	53	14.65	167.48	47.15	54.202	31.243
43102	TNMOC	32	28.03	348.39	60.68	79.506	66.028

In 2016, the Office of Air Quality Monitoring (AQM) of the Department of Environmental Quality (DEQ) operated an Air Toxics Monitoring Network (ATMN). The ATMN consists of three separate monitoring programs. The Urban Air Toxics Monitoring Program (UATM), The National Air Toxics Trend Stations Program (NATTS), and The Community Air Toxics Assessment Monitoring Program (CAMP).

The UATM program consisted of three sites that were located at: the Carter G. Woodson Middle School in Hopewell; DEQ Tidewater Regional Office (TRO) in Virginia Beach; and Lee District Park in Fairfax County. Sampling at these sites consisted of Volatile Organic Compounds (VOC), Carbonyls, and Total Suspended Particulate (TSP) Metals. Each of the UATM sites had a sampling schedule consisting of 24-hour samples collected every 6th day. Data from these sites will be used to characterize air toxics concentrations in the respective urban areas.

AQM used the manual method for collecting ambient air samples for VOC analysis. Whole air samples were collected using evacuated Silco^T or SUMMA^T canisters and RMESI (RM Environmental Systems, Inc.) air samplers. Samples were analyzed by the Division of Consolidated Laboratory Services (DCLS), the Virginia state laboratory. DCLS used a Gas Chromatograph equipped with a Mass Selective Detector and employed method TO15. Carbonyls were collected on DNPH (2,4-Dinitrophenylhydrazine) treated sorbent tubes using ATEC 8000 cartridge samplers. Samples taken were analyzed by the DCLS, using a Liquid Chromatographic procedure designated as method TO11A. Metals samples were collected using a high volume Total Suspended Particulate (TSP) sampler and were analyzed by the DCLS. Analysis utilized inductively coupled plasma mass spectrometry (ICP-MS) using method IO-3.1 and IO-3.5.

The NATTS program operated one station located at the MathScience Innovation Center (MSIC) in Henrico County. The NATTS site had a sampling schedule consisting of 24-hour samples collected every 6th day. Data from this site will be evaluated along with data from all of the NATTS sites nationally. AQM used the manual method for collecting ambient air samples for VOC analysis. Whole air samples were collected using evacuated Silco^T canisters and RMESI (RM Environmental Systems, Inc.) air samplers. Each sample was analyzed by the DCLS, using a Gas Chromatograph equipped with a Mass Selective Detector, utilizing method TO15. Carbonyls were collected on DNPH (2,4-Dinitrophenylhydrazine) treated sorbent tubes using ATEC 8000 cartridge samplers. Samples were analyzed by DCLS using a Liquid Chromatographic procedure, and the TO11A method. The Metals samples were collected using a high volume 10 micron Particulate Matter (PM10) sampler and analyzed by the DCLS. Analysis utilized Inductively Coupled Plasma Mass Spectrometry (ICP-MS) using method IO-3.1 and IO-3.5.

Detectable VOC in 24-Hour Canister Samples
GC/MSD - MathScience Innovation Center (NATTS Site), Henrico County, VA
January 1 to December 31, 2016- Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
43205	Propylene	58	0.00	2.00	0.00	0.034	0.263
43207	Freon 113	58	0.10	0.70	0.20	0.172	0.087
43208	Freon 114	58	0.00	0.00	0.00	0.000	0.000
43209	Ethyl Acetate	58	0.00	1.50	0.00	0.093	0.282
43218	1,3-Butadiene	58	0.00	0.00	0.00	0.000	0.000
43231	Hexane	58	0.00	3.40	0.60	0.750	0.615
43232	Heptane	58	0.00	1.70	0.40	0.359	0.400
43248	Cyclohexane	58	0.00	0.40	0.00	0.036	0.097
43372	MTBE	58	0.00	0.00	0.00	0.000	0.000
43441	Methyl Methacrylate	58	0.00	0.00	0.00	0.000	0.000
43505	Acrolein	58	0.00	10.00	0.80	1.041	1.607
43702	Acetonitrile	58	1.80	81.20	9.35	15.047	16.863
43704	Acrylonitrile	58	0.00	0.00	0.00	0.000	0.000
43801	Chloromethane	58	0.40	0.80	0.60	0.579	0.089
43802	Dichloromethane	58	0.10	9.60	0.75	1.326	1.623
43803	Chloroform	58	0.00	0.10	0.00	0.002	0.013
43804	Carbon Tetrachloride	58	0.10	0.10	0.10	0.100	0.000
43806	Bromoform (Tribromomethane)	58	0.00	0.00	0.00	0.000	0.000
43811	Trichlorofluoromethane	58	0.20	0.40	0.30	0.272	0.056
43812	Chloroethane	58	0.00	0.00	0.00	0.000	0.000
43813	1,1-Dichloroethane	58	0.00	0.00	0.00	0.000	0.000
43814	Methyl chloroform	58	0.00	0.00	0.00	0.000	0.000
43815	Ethylene dichloride	58	0.00	0.00	0.00	0.000	0.000
43817	Tetrachloroethylene	58	0.00	0.10	0.00	0.028	0.045
43818	1,1,2,2-Tetrachloroethane	58	0.00	0.00	0.00	0.000	0.000
43819	Bromomethane	58	0.00	0.00	0.00	0.000	0.000
43820	1,1,2-Trichloroethane	58	0.00	0.00	0.00	0.000	0.000
43823	Dichlorodifluoromethane	58	0.40	0.70	0.50	0.514	0.074
43824	Trichloroethylene	58	0.00	0.00	0.00	0.000	0.000
43826	1,1-Dichloroethylene	58	0.00	0.00	0.00	0.000	0.000
43828	Bromodichloromethane	58	0.00	0.00	0.00	0.000	0.000
43829	1,2-Dichloropropane	58	0.00	0.00	0.00	0.000	0.000
43830	trans-1,3-Dichlopropylene	58	0.00	0.00	0.00	0.000	0.000
43831	cis-1,3-Dichlopropylene	58	0.00	0.00	0.00	0.000	0.000
43832	Dibromochloromethane	58	0.00	0.00	0.00	0.000	0.000
43838	Trans-1,2-Dichloroethene	58	0.00	0.00	0.00	0.000	0.000
43839	cis-1,2-Dichloroethene	58	0.00	0.00	0.00	0.000	0.000
43843	Ethylene Dibromide	58	0.00	0.10	0.00	0.002	0.013
43844	Hexachlorobutadiene	58	0.00	0.10	0.00	0.002	0.013
43860	Vinyl Chloride	58	0.00	0.00	0.00	0.000	0.000
45109	m/p-Xylene	58	0.20	3.60	0.60	0.767	0.616
45201	Benzene	58	0.00	3.20	0.80	0.976	0.698
45202	Toluene	58	0.60	7.60	1.50	1.831	1.316
45203	Ethylbenzene	58	0.20	1.60	0.60	0.643	0.349
45204	o-Xylene	58	0.00	1.50	0.20	0.329	0.260
45207	1,3,5-Trimethylbenzene	58	0.00	0.50	0.00	0.067	0.121
45208	1,2,4-Trimethylbenzene	58	0.00	1.60	0.30	0.372	0.323
45213	p-Ethyltoluene	58	0.00	0.50	0.00	0.090	0.131
45220	Styrene	58	0.00	2.90	0.60	0.824	0.615
45801	Chlorobenzene	58	0.00	0.10	0.00	0.002	0.013
45805	1,2-Dichlorobenzene	58	0.00	0.70	0.00	0.055	0.116
45806	1,3-Dichlorobenzene	58	0.00	1.00	0.00	0.074	0.162
45807	1,4-Dichlorobenzene	58	0.00	1.20	0.10	0.112	0.197
45810	1,2,4-Trichlorobenzene	58	0.00	1.00	0.00	0.074	0.200
46401	Tetrahydrofuran	58	0.00	1.00	0.00	0.074	0.216

Detectable VOC in 24-Hour Canister Samples
GC/MSD - Carter G. Woodson Middle School (UATM Site), Hopewell, VA
January 1 to December 31, 2016 - Concentrations are in ppbV

(NonDetects are considered zeros for statistical purposes. Results below MDLs and/or Reporting Limits are reported.)

Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
43205	Propylene	58	0.00	0.79	0.00	0.052	0.164
43207	Freon 113	58	0.03	0.21	0.08	0.084	0.025
43208	Freon 114	58	0.00	0.00	0.00	0.000	0.000
43209	Ethyl Acetate	58	0.00	0.19	0.00	0.012	0.033
43218	1,3-Butadiene	58	0.00	0.00	0.00	0.000	0.000
43231	N-Hexane	58	0.00	0.57	0.08	0.094	0.103
43232	N-Heptane	58	0.00	0.36	0.04	0.055	0.074
43248	Cyclohexane	58	0.00	0.08	0.00	0.005	0.015
43372	Methyl Tert-Butyl Ether	58	0.00	0.00	0.00	0.000	0.000
43441	Methyl Methacrylate	58	0.00	0.00	0.00	0.000	0.000
43505	Acrolein	58	0.00	3.21	0.45	0.595	0.566
43702	Acetonitrile	58	0.00	7.01	1.94	2.545	2.025
43704	Acrylonitrile	58	0.00	0.00	0.00	0.000	0.000
43801	Chloromethane	58	0.19	0.75	0.58	0.571	0.081
43802	Dichloromethane	58	0.02	0.34	0.08	0.088	0.046
43803	Chloroform	58	0.00	0.04	0.02	0.021	0.011
43804	Carbon Tetrachloride	58	0.01	0.10	0.07	0.068	0.016
43806	Bromoform	58	0.00	0.00	0.00	0.000	0.000
43811	Trichlorofluoromethane	58	0.07	0.33	0.25	0.252	0.034
43812	Ethyl Chloride	58	0.00	0.53	0.00	0.020	0.084
43813	1,1-Dichloroethane	58	0.00	0.00	0.00	0.000	0.000
43814	Methyl Chloroform	58	0.00	0.00	0.00	0.000	0.000
43815	1,2-Dichloroethane	58	0.00	0.00	0.00	0.000	0.000
43817	Tetrachloroethylene	58	0.00	0.05	0.00	0.009	0.011
43818	1,1,2,2-Tetrachloroethane	58	0.00	0.01	0.00	0.000	0.001
43819	Bromomethane	58	0.00	0.00	0.00	0.000	0.000
43820	1,1,2-Trichloroethane	58	0.00	0.00	0.00	0.000	0.000
43823	Dichlorodifluoromethane	58	0.15	0.62	0.49	0.495	0.066
43824	Trichloroethylene	58	0.00	0.02	0.00	0.001	0.003
43826	1,1-Dichloroethene	58	0.00	0.00	0.00	0.000	0.000
43828	Bromodichloromethane	58	0.00	0.00	0.00	0.000	0.000
43829	1,2-Dichloropropane	58	0.00	0.00	0.00	0.000	0.000
43830	trans-1,3-Dichloropropylene	58	0.00	0.00	0.00	0.000	0.000
43831	Cis-1,3-Dichloropropylene	58	0.00	0.00	0.00	0.000	0.000
43832	Dibromochloromethane	58	0.00	0.00	0.00	0.000	0.000
43838	trans-1,2-Dichloroethene	58	0.00	0.00	0.00	0.000	0.000
43839	cis-1,2-Dichloroethene	58	0.00	0.00	0.00	0.000	0.000
43843	1,2-Dibromoethane	58	0.00	0.01	0.00	0.000	0.002
43844	Hexachloro-1,3-Butadiene	58	0.00	0.01	0.00	0.000	0.001
43860	Vinyl Chloride	58	0.00	0.00	0.00	0.000	0.000
45109	m & p- Xylene	58	0.00	0.42	0.04	0.067	0.075
45201	Benzene	58	0.04	0.64	0.11	0.144	0.106
45202	Toluene	58	0.03	0.80	0.14	0.175	0.150
45203	Ethylbenzene	58	0.00	0.13	0.02	0.025	0.028
45204	o-Xylene	58	0.00	0.18	0.02	0.030	0.031
45207	1,3,5-Trimethylbenzene	58	0.00	0.06	0.00	0.005	0.012
45208	1,2,4-Trimethylbenzene	58	0.00	0.18	0.02	0.032	0.036
45213	4-Ethyltoluene	58	0.00	0.18	0.00	0.008	0.027
45220	Styrene	58	0.00	0.56	0.00	0.025	0.080
45801	Chlorobenzene	58	0.00	0.00	0.00	0.000	0.000
45805	1,2-Dichlorobenzene	58	0.00	0.06	0.00	0.003	0.010
45806	1,3-Dichlorobenzene	58	0.00	0.07	0.00	0.004	0.012
45807	1,4-Dichlorobenzene	58	0.00	0.12	0.00	0.009	0.024
45810	1,2,4-Trichlorobenzene	58	0.00	0.17	0.00	0.007	0.030
46401	Tetrahydrofuran	58	0.00	0.15	0.00	0.009	0.029

Detectable VOC in 24-Hour Canister Samples
GC/MSD - Tidewater Regional Office (UATM Site), Va. Beach, VA
January 1 to December 31, 2016 – Concentrations are in ppbV

(NonDetects are considered zeros for statistical purposes. Results below MDLs and/or Reporting Limits are reported.)

Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
43205	Propylene	50	0.00	1.90	0.00	0.085	0.313
43207	Freon 113	50	0.08	0.09	0.08	0.074	0.011
43208	Freon 114	50	0.00	0.00	0.00	0.000	0.000
43209	Ethyl Acetate	50	0.00	0.00	0.00	0.000	0.000
43218	1,3-Butadiene	50	0.00	0.00	0.00	0.000	0.000
43231	N-Hexane	50	0.19	0.59	0.11	0.118	0.119
43232	N-Heptane	50	0.08	0.25	0.05	0.053	0.059
43248	Cyclohexane	50	0.00	0.07	0.00	0.006	0.016
43372	Methyl Tert-Butyl Ether	50	0.00	0.00	0.00	0.000	0.000
43441	Methyl Methacrylate	50	0.00	0.00	0.00	0.000	0.000
43505	Acrolein	50	1.30	1.64	0.46	0.529	0.417
43702	Acetonitrile	50	0.67	4.09	0.50	0.553	0.607
43704	Acrylonitrile	50	0.00	0.00	0.00	0.000	0.000
43801	Chloromethane	50	0.62	0.76	0.59	0.585	0.097
43802	Dichloromethane	50	0.10	0.20	0.08	0.088	0.031
43803	Chloroform	50	0.03	0.05	0.02	0.024	0.010
43804	Carbon Tetrachloride	50	0.07	0.11	0.07	0.069	0.017
43806	Bromoform	50	0.00	0.00	0.00	0.000	0.000
43811	Trichlorofluoromethane	50	0.28	0.32	0.25	0.252	0.039
43812	Ethyl Chloride	50	0.00	0.00	0.00	0.000	0.000
43813	1,1-Dichloroethane	50	0.00	0.00	0.00	0.000	0.000
43814	Methyl Chloroform	50	0.00	0.00	0.00	0.000	0.000
43815	1,2-Dichloroethane	50	0.00	0.00	0.00	0.000	0.000
43817	Tetrachloroethylene	50	0.04	1.19	0.04	0.111	0.211
43818	1,1,2,2-Tetrachloroethane	50	0.00	0.00	0.00	0.000	0.000
43819	Bromomethane	50	0.00	0.00	0.00	0.000	0.000
43820	1,1,2-Trichloroethane	50	0.00	0.00	0.00	0.000	0.000
43823	Dichlorodifluoromethane	50	0.51	0.62	0.49	0.491	0.076
43824	Trichloroethylene	50	0.00	0.04	0.00	0.002	0.007
43826	1,1-Dichloroethene	50	0.00	0.00	0.00	0.000	0.000
43828	Bromodichloromethane	50	0.00	0.00	0.00	0.000	0.000
43829	1,2-Dichloropropane	50	0.00	0.00	0.00	0.000	0.000
43830	trans-1,3-Dichloropropylene	50	0.00	0.00	0.00	0.000	0.000
43831	Cis-1,3-Dichloropropylene	50	0.00	0.00	0.00	0.000	0.000
43832	Dibromochloromethane	50	0.00	0.00	0.00	0.000	0.000
43838	trans-1,2-Dichloroethene	50	0.00	0.00	0.00	0.000	0.000
43839	cis-1,2-Dichloroethene	50	0.00	0.00	0.00	0.000	0.000
43843	1,2-Dibromoethane	50	0.02	0.02	0.00	0.000	0.003
43844	Hexachloro-1,3-Butadiene	50	0.00	0.00	0.00	0.000	0.000
43860	Vinyl Chloride	50	0.00	0.00	0.00	0.000	0.000
45109	m & p- Xylene	50	0.11	0.50	0.05	0.087	0.088
45201	Benzene	50	0.23	0.48	0.12	0.148	0.110
45202	Toluene	50	0.41	1.41	0.18	0.260	0.241
45203	Ethylbenzene	50	0.05	0.15	0.02	0.026	0.029
45204	o-Xylene	50	0.05	0.20	0.03	0.037	0.035
45207	1,3,5-Trimethylbenzene	50	0.01	0.05	0.00	0.006	0.010
45208	1,2,4-Trimethylbenzene	50	0.03	0.19	0.02	0.033	0.032
45213	4-Ethyltoluene	50	0.00	0.06	0.00	0.006	0.012
45220	Styrene	50	0.00	0.06	0.00	0.010	0.016
45801	Chlorobenzene	50	0.00	0.00	0.00	0.000	0.000
45805	1,2-Dichlorobenzene	50	0.07	0.07	0.00	0.004	0.013
45806	1,3-Dichlorobenzene	50	0.08	0.08	0.00	0.006	0.016
45807	1,4-Dichlorobenzene	50	0.11	0.11	0.00	0.009	0.022
45810	1,2,4-Trichlorobenzene	50	0.00	0.15	0.00	0.008	0.030
46401	Tetrahydrofuran	50	0.00	0.24	0.00	0.006	0.035

Detectable VOC in 24-Hour Canister Samples
GC/MSD - Lee District Park (UATM Site), Fairfax County, VA
January 1 to December 31, 2016 - Concentrations are in ppbV

(NonDetects are considered zeros for statistical purposes. Results below MDLs and/or Reporting Limits are reported.)

Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
43205	Propylene	59	0.00	1.92	0.00	0.114	0.322
43207	Freon 113	59	0.06	0.09	0.07	0.075	0.006
43208	Freon 114	59	0.00	0.00	0.00	0.000	0.000
43209	Ethyl Acetate	59	0.00	6.47	0.00	0.110	0.842
43218	1,3-Butadiene	59	0.00	0.00	0.00	0.000	0.000
43231	N-Hexane	59	0.00	0.33	0.09	0.083	0.066
43232	N-Heptane	59	0.00	0.12	0.00	0.031	0.037
43248	Cyclohexane	59	0.00	0.04	0.00	0.003	0.009
43372	Methyl Tert-Butyl Ether	59	0.00	0.00	0.00	0.000	0.000
43441	Methyl Methacrylate	59	0.00	0.00	0.00	0.000	0.000
43505	Acrolein	59	0.00	4.81	0.91	1.200	1.111
43702	Acetonitrile	59	0.00	226.42	94.14	81.207	72.402
43704	Acrylonitrile	59	0.00	0.00	0.00	0.000	0.000
43801	Chloromethane	59	0.00	0.79	0.56	0.570	0.107
43802	Dichloromethane	59	0.06	1.80	0.09	0.123	0.223
43803	Chloroform	59	0.00	0.04	0.02	0.025	0.010
43804	Carbon Tetrachloride	59	0.03	0.12	0.06	0.064	0.018
43806	Bromoform	59	0.00	0.00	0.00	0.000	0.000
43811	Trichlorofluoromethane	59	0.21	0.44	0.24	0.253	0.035
43812	Ethyl Chloride	59	0.00	0.00	0.00	0.000	0.000
43813	1,1-Dichloroethane	59	0.00	0.00	0.00	0.000	0.000
43814	Methyl Chloroform	59	0.00	0.00	0.00	0.000	0.000
43815	1,2-Dichloroethane	59	0.00	0.00	0.00	0.000	0.000
43817	Tetrachloroethylene	59	0.00	0.08	0.02	0.025	0.018
43818	1,1,2,2-Tetrachloroethane	59	0.00	0.01	0.00	0.000	0.001
43819	Bromomethane	59	0.00	0.00	0.00	0.000	0.000
43820	1,1,2-Trichloroethane	59	0.00	0.00	0.00	0.000	0.000
43823	Dichlorodifluoromethane	59	0.41	0.64	0.48	0.492	0.052
43824	Trichloroethylene	59	0.00	0.12	0.00	0.010	0.025
43826	1,1-Dichloroethene	59	0.00	0.00	0.00	0.000	0.000
43828	Bromodichloromethane	59	0.00	0.00	0.00	0.000	0.000
43829	1,2-Dichloropropane	59	0.00	0.00	0.00	0.000	0.000
43830	trans-1,3-Dichloropropylene	59	0.00	0.00	0.00	0.000	0.000
43831	Cis-1,3-Dichloropropylene	59	0.00	0.00	0.00	0.000	0.000
43832	Dibromochloromethane	59	0.00	0.00	0.00	0.000	0.000
43838	trans-1,2-Dichloroethene	59	0.00	0.00	0.00	0.000	0.000
43839	cis-1,2-Dichloroethene	59	0.00	0.00	0.00	0.000	0.000
43843	1,2-Dibromoethane	59	0.00	0.01	0.00	0.000	0.001
43844	Hexachloro-1,3-Butadiene	59	0.00	0.01	0.00	0.000	0.001
43860	Vinyl Chloride	59	0.00	0.00	0.00	0.000	0.000
45109	m & p- Xylene	59	0.02	0.60	0.08	0.103	0.091
45201	Benzene	59	0.05	0.52	0.12	0.133	0.078
45202	Toluene	59	0.04	0.83	0.24	0.276	0.180
45203	Ethylbenzene	59	0.00	0.16	0.03	0.038	0.031
45204	o-Xylene	59	0.00	0.19	0.04	0.045	0.032
45207	1,3,5-Trimethylbenzene	59	0.00	0.02	0.00	0.003	0.006
45208	1,2,4-Trimethylbenzene	59	0.00	0.06	0.02	0.023	0.014
45213	4-Ethyltoluene	59	0.00	0.05	0.00	0.003	0.008
45220	Styrene	59	0.00	0.07	0.02	0.023	0.023
45801	Chlorobenzene	59	0.00	0.00	0.00	0.000	0.000
45805	1,2-Dichlorobenzene	59	0.00	0.02	0.00	0.002	0.005
45806	1,3-Dichlorobenzene	59	0.00	0.02	0.00	0.002	0.005
45807	1,4-Dichlorobenzene	59	0.00	0.02	0.00	0.005	0.007
45810	1,2,4-Trichlorobenzene	59	0.00	0.04	0.00	0.002	0.008
46401	Tetrahydrofuran	59	0.00	0.23	0.00	0.007	0.034

24 Hour Carbonyl Sampling 2016 Summary Statistical Analysis

Concentrations are in $\mu\text{g}/\text{m}^3$
(non detects are counted as zeros for statistical purposes)

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
Carter G. Woodson Middle School	43502	Formaldehyde	58	0.88	7.51	2.04	2.489	1.322
	43503	Acetaldehyde	58	0.53	3.43	1.44	1.549	0.654
	43504	Propionaldehyde	58	0.00	0.53	0.20	0.205	0.086
	43551	Acetone	58	0.42	14.80	3.61	3.899	2.378
	43552	Methyl Ethyl Ketone	58	0.10	1.03	0.42	0.423	0.189
	43560	Methyl Isobutyl Ketone	58	0.00	0.00	0.00	0.000	0.000

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
Tidewater Regional Office	43502	Formaldehyde	59	0.45	4.46	1.80	1.926	0.842
	43503	Acetaldehyde	59	0.23	2.74	1.11	1.110	0.476
	43504	Propionaldehyde	59	0.00	0.61	0.18	0.188	0.099
	43551	Acetone	59	0.00	7.58	2.49	2.685	1.898
	43552	Methyl Ethyl Ketone	59	0.00	1.04	0.35	0.344	0.214
	43560	Methyl Isobutyl Ketone	59	0.00	0.25	0.00	0.015	0.053

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
Lee Park	43502	Formaldehyde	60	0.84	7.13	2.00	2.420	1.330
	43503	Acetaldehyde	60	0.54	3.32	1.20	1.278	0.520
	43504	Propionaldehyde	60	0.00	0.51	0.22	0.234	0.107
	43551	Acetone	60	1.43	6.58	3.49	3.497	1.227
	43552	Methyl Ethyl Ketone	60	0.17	1.09	0.45	0.476	0.184
	43560	Methyl Isobutyl Ketone	60	0.00	0.04	0.00	0.001	0.005

NATTS Carbonyl Sampling 2016 Summary Statistical Analysis

Concentrations are in $\mu\text{g}/\text{m}^3$
(non detects are counted as zeros for statistical purposes)

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
MathScience Innovation Center	43502	Formaldehyde	60	0.80	6.10	1.90	2.153	1.100
	43503	Acetaldehyde	60	0.70	3.30	1.40	1.555	0.570
	43504	Propionaldehyde	60	1.00	10.00	4.40	4.553	2.001
	43551	Acetone	60	0.00	0.70	0.30	0.305	0.117
	43552	Methyl Ethyl Ketone	60	0.20	1.40	0.60	0.660	0.268
	43560	Methyl Isobutyl Ketone	60	0.00	0.10	0.00	0.005	0.021

TSP Metals Sampling 2016 Summary Statistical Analysis

Concentrations are in ng/m³
(non detects are counted as zeros for statistical purposes)

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
Carter G. Woodson Middle School	12103	Arsenic	60	0.19	4.30	0.68	0.960	0.791
	12105	Beryllium	60	0.00	0.02	0.00	0.002	0.003
	12110	Cadmium	60	0.00	0.54	0.10	0.107	0.078
	12112	Chromium	60	1.31	4.86	1.86	1.904	0.493
	12128	Lead	60	0.61	6.11	1.92	2.128	0.995
	12132	Manganese	60	2.19	20.08	6.20	7.105	3.794
	12136	Nickel	60	0.30	1.79	0.59	0.662	0.253

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
Tidewater Regional Office	12103	Arsenic	61	0.22	8.95	0.85	1.339	1.449
	12105	Beryllium	61	0.00	0.01	0.00	0.002	0.003
	12110	Cadmium	61	0.00	0.16	0.05	0.055	0.039
	12112	Chromium	61	1.37	3.21	1.83	1.887	0.355
	12128	Lead	61	0.13	6.70	1.61	1.797	1.082
	12132	Manganese	61	0.41	20.23	4.50	5.547	3.788
	12136	Nickel	61	0.35	1.61	0.65	0.710	0.229

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
Lee Park	12103	Arsenic	61	0.15	2.73	0.78	0.867	0.595
	12105	Beryllium	61	0.00	0.02	0.00	0.002	0.003
	12110	Cadmium	61	0.00	0.69	0.07	0.092	0.093
	12112	Chromium	61	1.20	2.56	1.79	1.824	0.269
	12128	Lead	61	0.51	18.95	1.65	2.316	2.634
	12132	Manganese	61	1.67	26.26	5.05	5.816	3.820
	12136	Nickel	61	0.30	5.42	0.54	0.678	0.652

NATTS PM10 Metals Sampling 2016 Summary Statistical Analysis

Concentrations are in ng/m³
(non-detects and negative values are counted as zeros for statistical purposes)

Site	Parameter	Compound Name	Num	Minimum	Maximum	Median	Average	StDev
MathScience Innovation Center	82103	Arsenic	59	0.08	2.82	0.71	0.829	0.560
	82105	Beryllium	59	0.00	0.01	0.00	0.001	0.002
	82110	Cadmium	59	0.01	0.45	0.06	0.080	0.069
	82112	Chromium	59	1.59	2.45	1.83	1.854	0.172
	82128	Lead	59	0.67	6.58	1.72	2.014	1.108
	82132	Manganese	59	0.67	6.87	2.01	2.259	1.161
	82136	Nickel	59	0.32	1.14	0.47	0.505	0.146

AQI (Air Quality Index)



What is the AQI?

The air quality index (AQI) is a measurement designed to indicate how clean the air is in an area, and it also provides information about health effects associated with air pollution. The index is reported daily, or in some cases continuously, and calculated from measured concentrations of five major pollutants regulated by the Clean Air Act: ground-level ozone, particulate matter, carbon monoxide, sulfur dioxide, and nitrogen dioxide. EPA has established national ambient air quality standards (NAAQS) for each of these pollutants to protect public health, and the index is derived from the NAAQS. State and local agencies are required to report the AQI in areas where the population is 350,000 or more, although it is often reported in additional areas as a public service.

How does the AQI work?

The AQI range is from 0 to 500, with the low numbers representing good air quality and the high numbers indicating unhealthy or even hazardous air quality. The index is divided into six categories with coordinating color codes. In addition, each category has a health-related message associated with it, to inform the public of possible health effects that may arise as a result of breathing polluted air.

Generally, an index of 100 corresponds to the national air quality standard for the pollutant, which is the level that EPA has established to protect public health. Levels below 100 are considered satisfactory, while numbers above 100 are considered unhealthy, first for sensitive groups, and then for the general public as the index value increases.

How is the AQI calculated?

The AQI is calculated from air pollution measurements collected at monitoring sites across the country. The reporting agency must calculate an index for each pollutant from the measured concentrations at all monitoring sites in an area using a standard formula developed by EPA. The pollutant with the highest index is reported as the "primary pollutant", and the highest index is reported as the AQI for the area. If the AQI is above 100, then the agency must report which groups may be sensitive to the primary pollutant. If two or more pollutants have indexes above 100, then the agency must report all groups that may be affected by those pollutants.

How do I find the AQI for my area?

DEQ reports the air quality index for Roanoke, Winchester, Richmond, Hampton Roads, and Northern Virginia on the internet at <http://vadeq.tx.sutron.com>. Air quality forecasts and current air quality data can be obtained at the DEQ site, as well as links to other air quality websites. EPA also reports air quality conditions for the United States at www.airnow.gov.

In addition to the internet, current and forecasted AQI levels are broadcast on local television and radio weather reports in many areas, as well as printed in newspapers. By reaching out to the public using these different media, individuals can plan their activities to reduce exposure during episodes of poor air quality, and they can also take steps to reduce pollution.

For detailed information about the AQI, and on health effects of the pollutants that are included in the AQI, visit www.airnow.gov.

Air Quality Index (0-500)	Levels of Health Concern	Meaning
(0-50)	Good	Air quality is considered satisfactory, and air pollution poses little or no risk.
(51-100)	Moderate	Air quality is acceptable; however, for some pollutants there may be a moderate health concern for a very small number of people who are unusually sensitive to air pollution.
(101-150)	Unhealthy for Sensitive Groups	Members of sensitive groups may experience health effects. The general public is not likely to be affected.
(151-200)	Unhealthy	Everyone may begin to experience health effects; members of sensitive groups may experience more serious health effects.
(201-300)	Very Unhealthy	Health warnings of emergency conditions. The entire population is more likely to be affected.
(301-500)	Hazardous	Health alert: everyone may experience more serious health effects.

Note: Values above 500 are considered Beyond the AQI. Follow recommendations for the "Hazardous category." Additional information on reducing exposure to extremely high levels of particle pollution is available ["here"](#)

How OZONE IS FORMED



NO_x + VOCs + Heat & Sunlight = Ozone

Ground level or "bad" ozone is not emitted directly into the air, but is created by chemical reactions between oxides of nitrogen (NO_x) and volatile organic compounds (VOCs) in the presence of heat and sunlight.

Emissions from industrial facilities and electric utilities, motor vehicle exhaust, gasoline vapors and chemical solvents are some of the major sources of NO_x and VOCs.

For more information, please visit this site:

www.airnow.gov/index.cfm?action=resources.whatyoucando

Key Facts to Know About Ozone:

- Ozone in the air we breathe can cause serious health problems, including breathing difficulty, asthma attacks, lung damage, and early death.
- Ozone forms in the sun, usually on hot summer days. Ozone is worse in the afternoon and early evening, so plan outdoor activities for the morning.
- You can reduce your exposure to ozone and still get exercise! Use the Air Quality Index (AQI) at www.airnow.gov to plan your activity.

What is ozone?

Ozone is a colorless gas that can be good or bad, depending on where it is. Ozone in the stratosphere is good because it shields the earth from the sun's ultraviolet rays. Ozone at ground level, where we breathe, is bad because it can harm human health.

Ozone forms when two types of pollutants (VOCs and NOx) react in sunlight. These pollutants come from sources such as vehicles, industries, power plants, and products such as solvents and paints.

Why is ozone a problem?

Ozone can cause a number of health problems, including coughing, breathing difficulty, and lung damage. Exposure to ozone can make the lungs more susceptible to infection, aggravate lung diseases, increase the frequency of asthma attacks, and increase the risk of early death from heart or lung disease.

Do I need to be concerned?

Even healthy adults can experience ozone's harmful effects, but **some people may be at greater risk**. They include:

- People with lung disease such as asthma
- Children, including teenagers, because their lungs are still developing and they breathe more air per pound of body weight than adults
- Older adults
- People who are active outdoors, including outdoor workers

How can I protect myself?

Use the Air Quality Index (AQI) to plan outdoor activities. To keep the AQI handy, sign up for EnviroFlash emails, get the free AirNow app, or install the free widget on your website. Find all of these tools at www.airnow.gov.

Stay healthy: exercise, eat a balanced diet, and keep asthma under control with your asthma action plan.

When you see that the AQI is unhealthy, take simple steps to reduce your exposure:

- Choose a less-strenuous activity
- Take more breaks during outdoor activity
- Reschedule activities to the morning or to another day
- Move your activity inside where ozone levels are usually lower

Can I help reduce ozone?

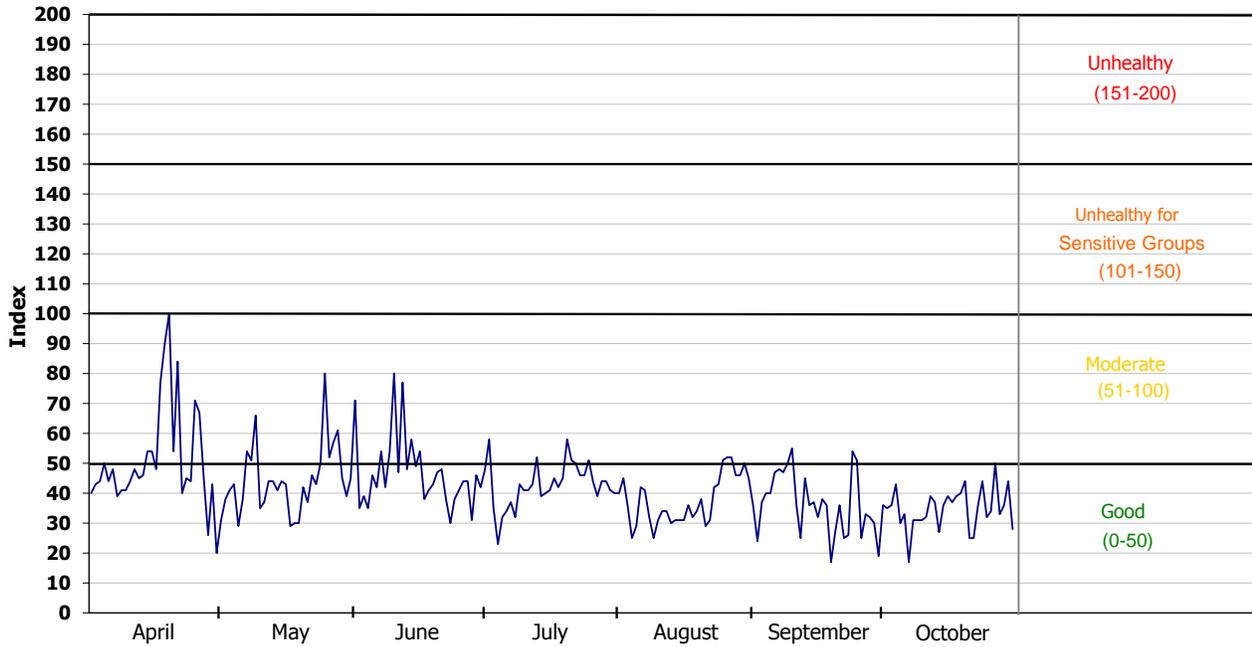
Yes! Here are a few tips.

- Turn off lights you are not using
- Drive less: carpool, use public transportation, bike or walk
- Keep your engine tuned, and don't let your engine idle
- When refueling: stop when the pump shuts off, avoid spilling fuel, and tighten your gas cap
- Inflate tires to the recommended pressure
- Use low-VOC paint and cleaning products, and seal and store them so they can't evaporate
- Watch for Air Quality Action Days in your area

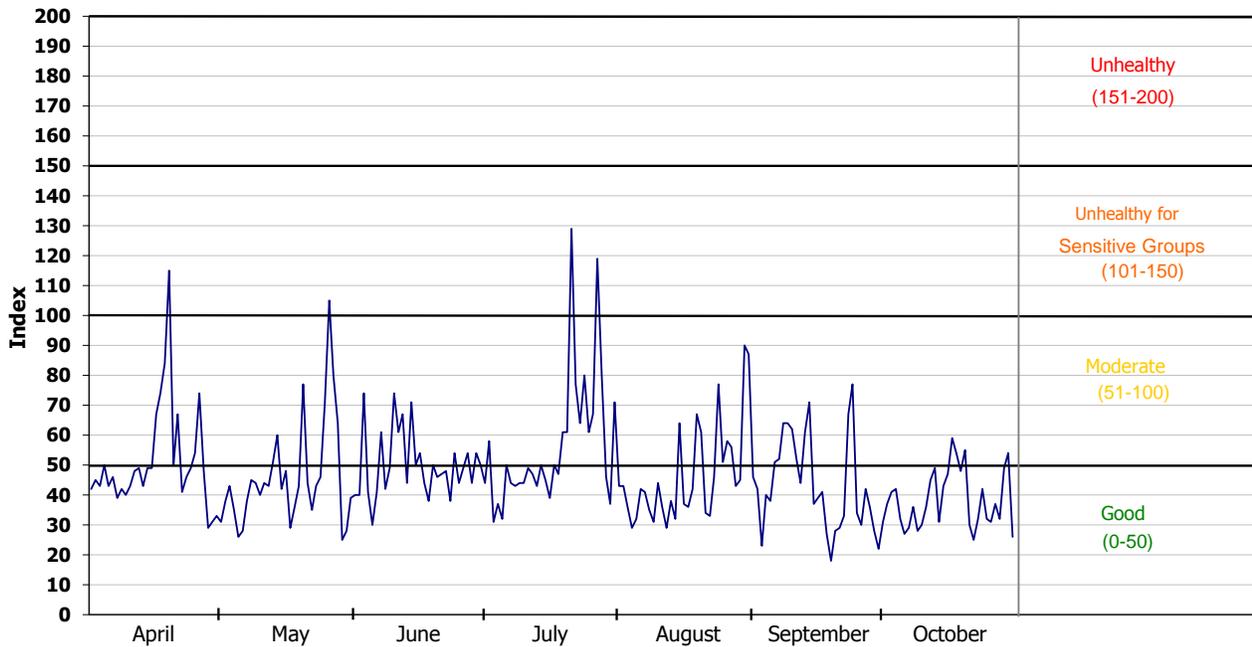


Office of Air Quality and Radiation EPA-456/F-15-002 www.airnow.gov

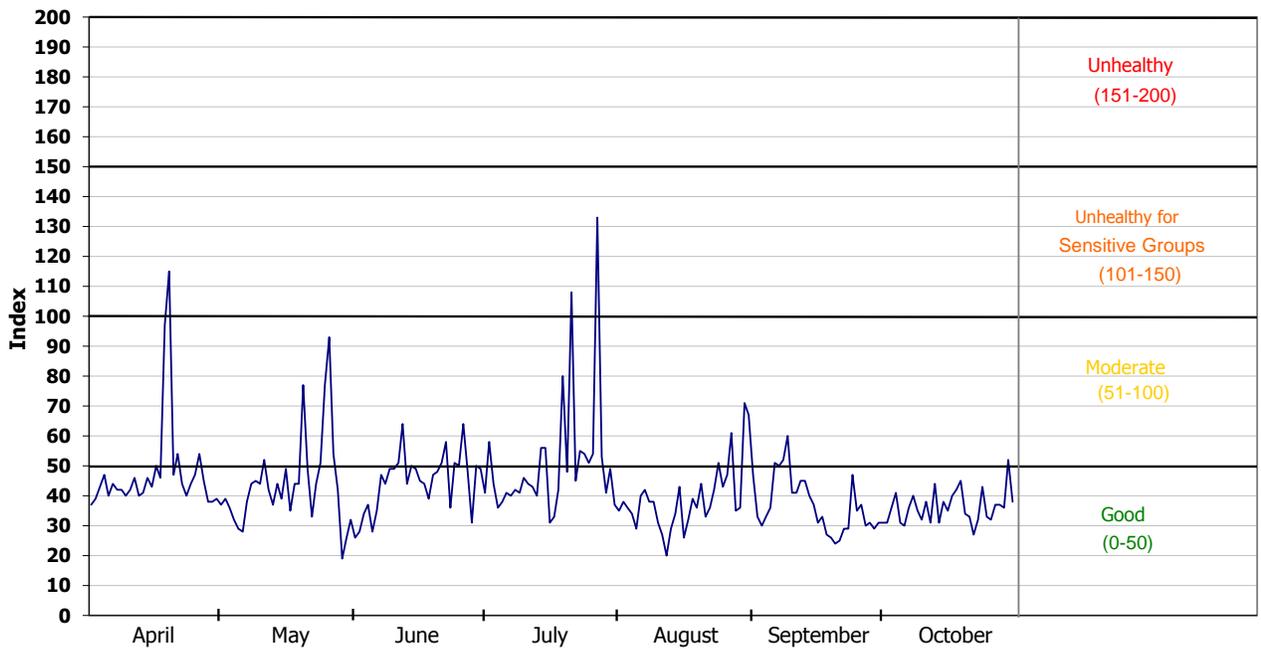
Ozone Air Quality Index Roanoke Area 2016



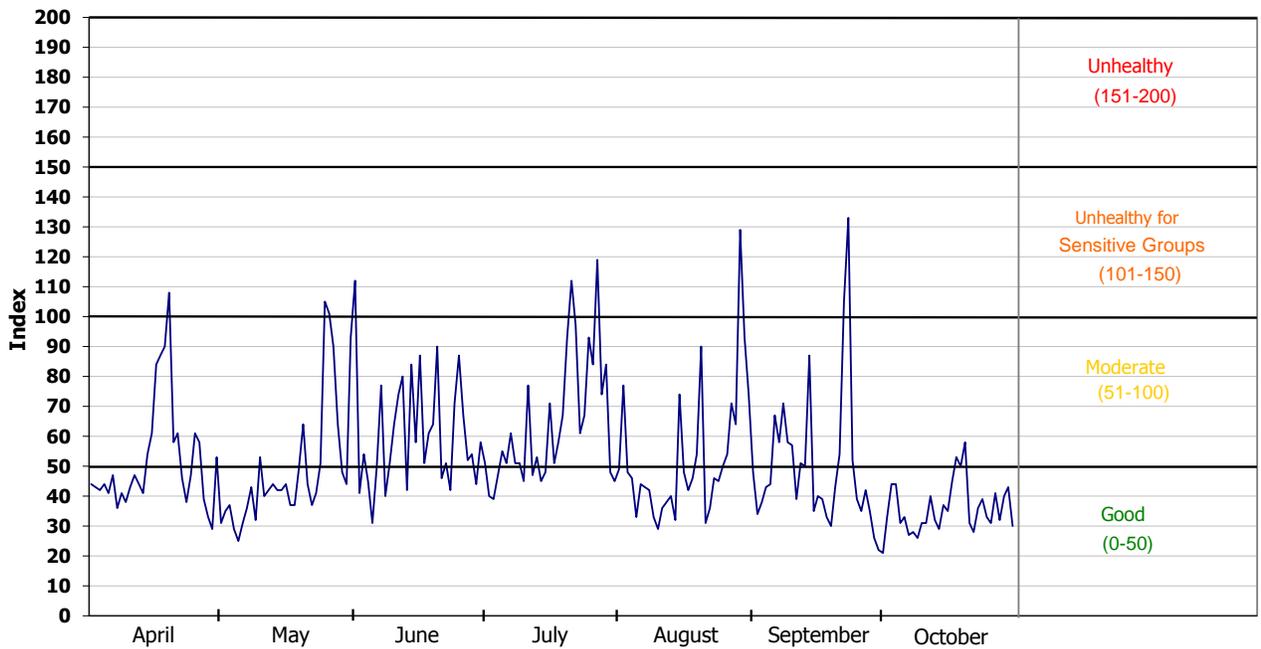
Ozone Air Quality Index Richmond - Petersburg Areas 2016



Ozone Air Quality Index Norfolk - Virginia Beach - Newport News Areas 2016



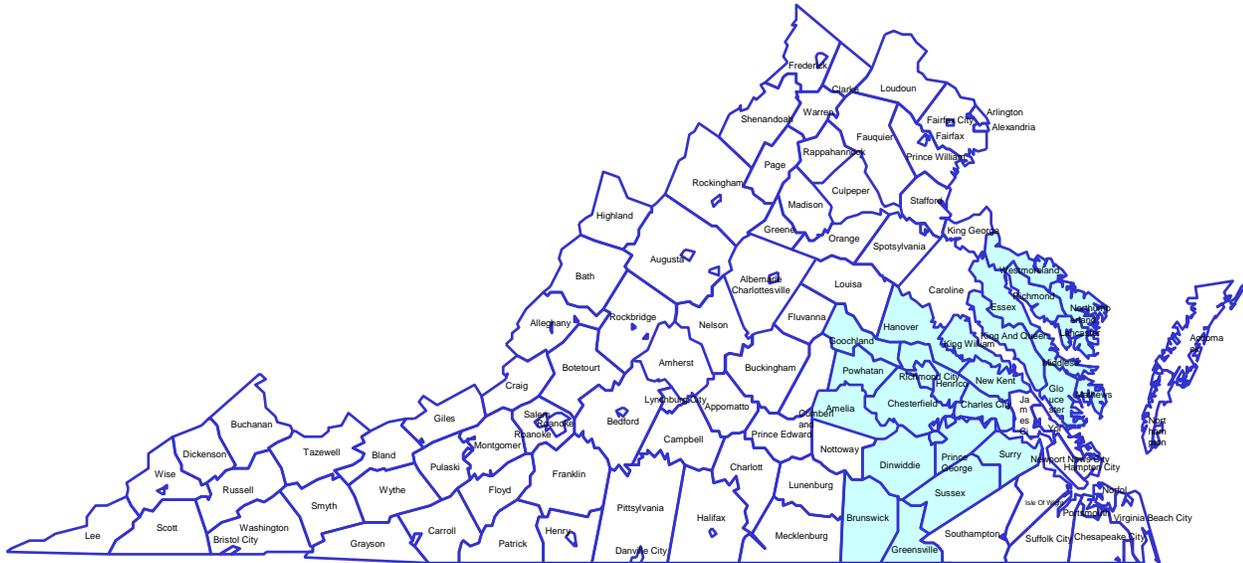
Ozone Air Quality Index Washington, DC Area 2016



Appendix A

AQM	Air Quality Monitoring
AQCR	Air Quality Control Region
ATMN	Air Toxics Monitoring Network
Avg.	Average
CAMP	Community Air Toxics Assessment Monitoring Program
CASTNET	Clean Air Status and Trends Network
CO	Carbon Monoxide
DEQ	Department of Environmental Quality
EPA	Environmental Protection Agency
IMPROVE	Interagency Monitoring of Protected Visual Environments
LAT	Latitude
LC	Local Conditions
LONG	Longitude
MARAMA	Mid-Atlantic Regional Air Management Association
MET.	Meteorological Instrumentation
MSA	Metropolitan Statistical Area
NA	Not Available
NAMS	National Air Monitoring Stations
NATTS	National Air Toxics Trend Stations
NMOC	Non-Methane Organic Compounds
NO ₂	Nitrogen Dioxide
NUM	Number of Samples
O ₃	Ozone
PAMHC	Total PAMS Hydrocarbon
PAMS	Photochemical Assessment Monitoring Station
PM ₁₀	Particulate Matter with an aerodynamic diameter less than or equal to 10 microns
PM _{2.5}	Particulate Matter with an aerodynamic diameter less than or equal to 2.5 microns
POLLUT.	Pollutant
ppbC	Part Per Billion of Carbon
ppbv	Part Per Billion by volume
ppm	Part Per Million
SESARM	Southeastern States Air Resource Managers, Inc.
SLAMS	State and Local Air Monitoring Station
SO ₂	Sulfur Dioxide
STD	Standard
STDEV	Standard Deviation
TEOM	Tapered Element Oscillating Microbalance (a method for continuously measuring PM _{2.5} in ambient air)
TNMOC	Total Nonmethane Organic Compound
UATM	Urban Air Toxics Monitoring Program
ug/m ³	Micrograms per cubic meter
VOC	Volatile Organic Compounds

Abbreviation Table



Piedmont Monitoring Network 2016

STATION NUMBER	POLLUT.	LOCATION	EPA ID	CITY/COUNTY	LAT/LONG
71-D	PM _{2.5}	Bensley Armory	51-041-0003	Chesterfield Co.	37.43467 -77.45118
71-H	O ₃	Beach Road Highway Shop	51-041-0004	Chesterfield Co.	37.35748 -77.59355
72-M	O ₃ , VOC, PM _{2.5} , PM ₁₀ , TEOM, Speciation, Toxics, Met, NCore, NATTS, PAMS	MathScience Innovation Center 2401 Hartman Street	51-087-0014	Henrico Co.	37.55652 -77.40027
72-N	PM _{2.5}	DEQ-Piedmont Regional Office 4949-A Cox Road	51-087-0015	Henrico Co.	37.67132 -77.56640
73-E	O ₃	McClellan Road	51-085-0003	Hanover Co.	37.60613 -77.21880
75-B	O ₃ , NO ₂ , SO ₂ , PM _{2.5}	Charles City County Route 608	51-036-0002	Charles City Co.	37.34438 -77.25925
154-M	PM ₁₀ , Toxics	Carter G. Woodson Middle School 1000 Winston Churchill Dr.	51-670-0010	Hopewell	37.28962 -77.29182
158-X	CO, NO ₂ , PM _{2.5} cont.	Joseph Bryan Park	51-760-0025	Richmond	37.59088 -77.46925

Contact Information for this Region:
 Piedmont Regional Office
 Jeffery Steers, Director
 4949-A Cox Road
 Glen Allen, VA 23060
 (804) 527-5053

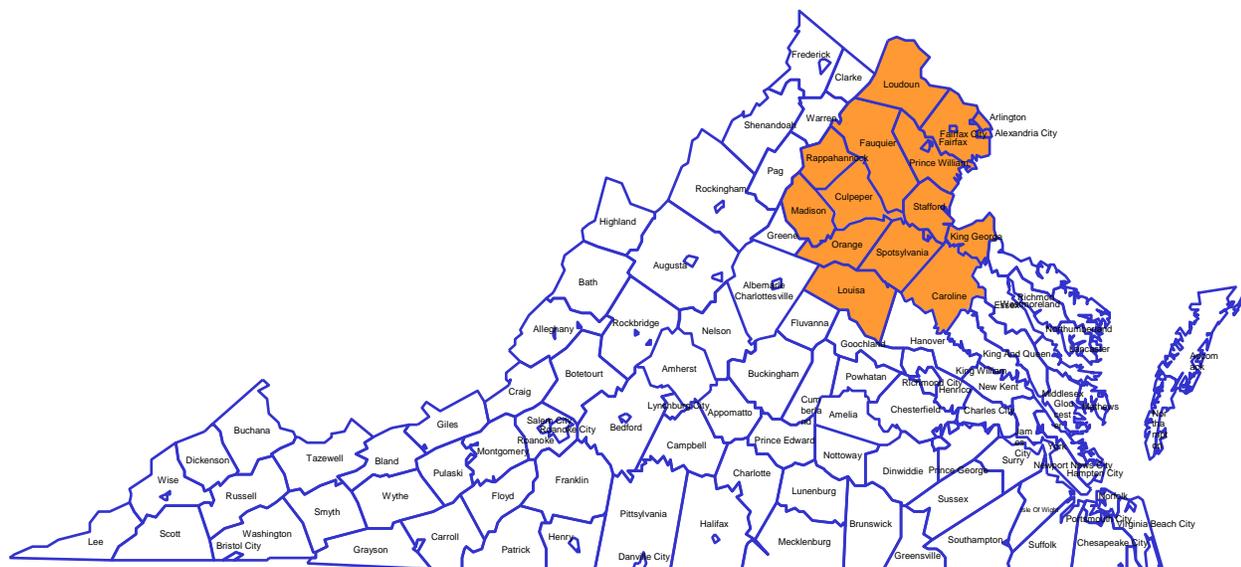


Tidewater Monitoring Network 2016

STATION NUMBER	POLLUT.	LOCATION	EPA ID	CITY/COUNTY	LAT/LONG
179-K	CO, SO ₂ , NO ₂ , O ₃ , PM _{2.5} , PM ₁₀ , TEOM	NASA Langley Research Center	51-650-0008	Hampton	37.10373 -76.38702
181-A1	CO, SO ₂ , NO ₂ , PM ₁₀ , PM _{2.5}	NOAA Property 2 nd and Woodis Avenue	51-710-0024	Norfolk	36.85555 -76.30135
183-E	O ₃	Tidewater Community College Frederick Campus	51-800-0004	Suffolk	36.90118 -76.43808
183-F	O ₃	Tidewater Research Station	51-800-0005	Suffolk	36.66525 -76.73078
184-J	PM _{2.5} , Toxics	DEQ – Tidewater Regional Office 5636 Southern Blvd.	51-810-0008	Va. Beach	36.84188 -76.18123

Contact information for this Region:

Craig Nicol, Director
5636 Southern Blvd.
Virginia Beach, VA 23462
(757) 518-2173



STATION NUMBER	POLLUT.	LOCATION	EPA ID	CITY/COUNTY	LAT/LONG
37-B	O ₃	Phelps Wildlife Area Route 651	51-061-0002	Sumerduck Fauquier Co.	38.47367 -77.76772
38-I	O ₃ , NO ₂ , PM _{2.5}	Broad Run High School Route 641	51-107-1005	Ashburn Loudoun Co.	39.02473 -77.48925
44-A	O ₃	Widewater Elementary School Den Rich Road	51-179-0001	Widewater Stafford Co.	38.48123 -77.37040
45-L	O ₃ , NO ₂	Long Park Route 15	51-153-0009	Prince William Co.	38.85287 -77.63462
46-B9	SO ₂ , O ₃ , PM _{2.5} , PM ₁₀ , TEOM, Toxics	Lee District Park Telegraph Road	51-059-0030	Franconia Fairfax Co.	38.77335 -77.10468
46-C2	CO, NO ₂ , PM _{2.5} cont.	6831 Backlick Road	51-059-0031	Springfield Fairfax Co.	38.76835 -77.18347
47-T	CO, NO ₂ , O ₃ , PM _{2.5}	Aurora Hills Visitors Center 18 th and Hayes Streets	51-013-0020	Arlington Co.	38.85770 -77.05922
48-A	O ₃	U.S.G.S. Geomagnetic Center	51-033-0001	Corbin Caroline Co.	38.20087 -77.37742
130-E	PM ₁₀	Hugh Mercer Elementary School 2100 Cowan Boulevard	51-630-0004	Fredericksburg	38.30225 -77.48712
L-126-H	PM ₁₀	435 Ferdinand Day Drive	51-510-0020	Alexandria	38.80493 -77.12687
N-35-A	O ₃ , TEOM, IMPROVE	Big Meadows, National Park Service	51-113-0003	Madison Co.	38.52280 -78.43487

Contact Information for this Region:
 Northern Regional Office
 Thomas Faha, Director
 13901 Crown Court
 Woodbridge, VA 22193
 (703) 583-3810

Data Capture Criteria

Minimum Number of Observations	
3-Hour Average	3 Consecutive Hourly Observations
8-Hour	6 Hourly Observations
24-Hour	18 Hourly Observations
Quarterly Averages & 3-month Averages	75% of Scheduled Samples
Yearly Averages	75% of Total Possible Observations
Annual Weighted Means	Four Complete Quarterly Averages

Pollutant		Primary/ Secondary	Averaging Time	Level	Form
Carbon Monoxide		Primary	8-hour	9 ppm	Not to be exceeded more than once per year
			1-hour	35 ppm	
Lead		Primary and Secondary	Rolling 3 month average	0.15 µg/m ³ (a)	Not to be exceeded
Nitrogen Dioxide		Primary	1-hour	100 ppb	98 th percentile of 1-hour daily maximum concentrations, averaged over 3 years
		Primary and Secondary	Annual	53 ppb (b)	Annual Mean
Ozone		Primary and Secondary	8-hour	0.070 ppm (c)	Annual fourth-highest daily maximum 8-hr concentration, averaged over 3 years
Particle Pollution	PM_{2.5}	Primary	Annual	12.0 µg/m ³	Annual Mean, averaged over 3 years
		Secondary	Annual	15.0 µg/m ³	Annual Mean, averaged over 3 years
		Primary and Secondary	24-hour	35 µg/m ³	98 th percentile, averaged over 3 years
	PM₁₀	Primary and Secondary	24-hour	150 µg/m ³	Not to be exceeded more than once per year on average over 3 years
Sulfur Dioxide		Primary	1-hour	75 ppb (d)	99 th percentile of 1-hour daily maximum concentrations, averaged over 3 years
		Secondary	3-hour	0.5 ppm	Not to be exceeded more than once per year

(a) In areas designated nonattainment for the Pb standards prior to the promulgation of the current (2008) standards, and for which implementation plans to attain or maintain the current (2008) standards have not been submitted and approved, the previous standards (1.5 µg/m³ as a calendar quarter average) also remain in effect.

(b) The level of the annual NO₂ standard is 0.053 ppm. It is shown here in terms of ppb for the purposes of clearer comparison to the 1-hour standard level.

(c) Final rule signed October 1, 2015, and effective December 28, 2015. The previous (2008) O₃ standards additionally remain in effect in some areas. Revocation of the previous (2008) O₃ standards and transitioning to the current (2015) standards will be addressed in the implementation rule for the current standards.

(d) The previous SO₂ standards (0.14 ppm 24-hour and 0.03 ppm annual) will additionally remain in effect in certain areas: (1) any area for which it is not yet 1 year since the effective date of designation under the current (2010) standards, and (2) any area for which implementation plans providing for attainment of the current (2010) standard have not been submitted and approved and which is designated nonattainment under the previous SO₂ standards or is not meeting the requirements of a SIP call under the previous SO₂ standards (40 CFR 50.4(3)). A SIP call is an EPA action requiring a state to resubmit all or part of its State Implementation Plan to demonstrate attainment of the required NAAQS.

Please see <https://www.epa.gov/criteria-air-pollutants/naaqs-table> for additional information concerning NAAQS.

National Ambient Air Quality Standards

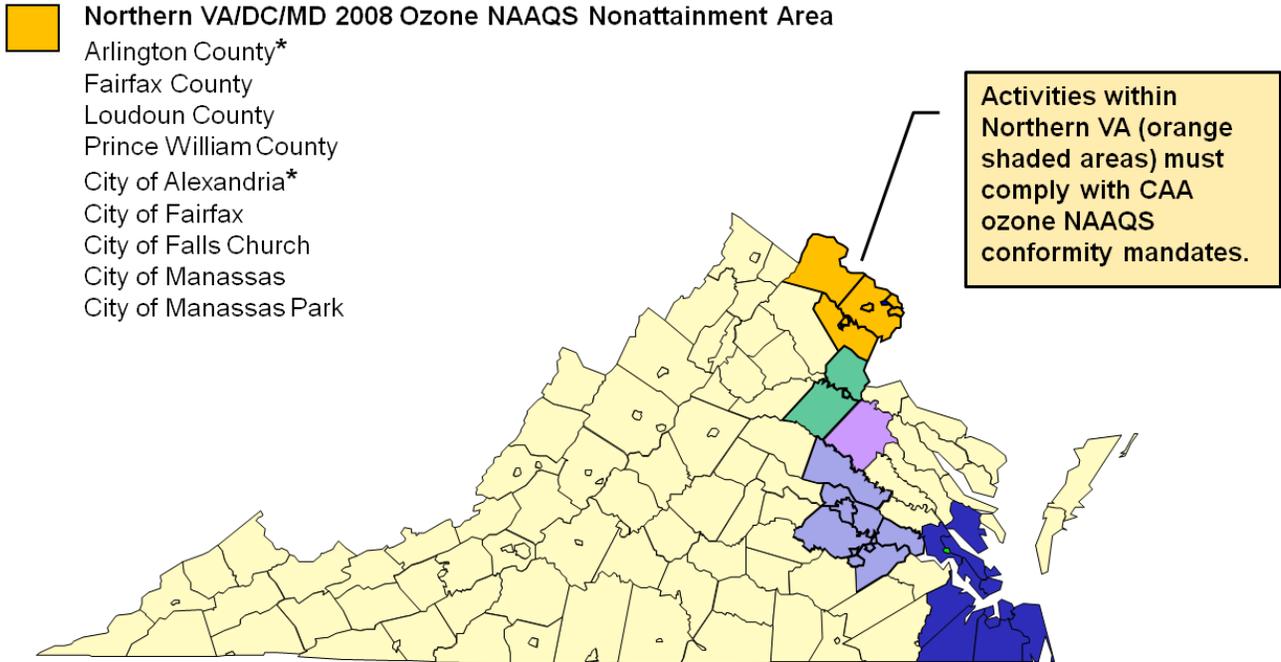
NCore/SLAMS 2016

REGION	PM _{2.5}	PM ₁₀	Pb	CO	SO ₂	NO ₂	O ₃	TOTAL
Southwest	1	1	---	---	---	---	1	3
Valley	3	1	---	---	1	1	4	10
Blue Ridge	3	---	2	1	1	1	1	9
Piedmont	4	2	---	2	2	3	4	17
Tidewater	3	2	---	2	2	2	3	14
*Northern	3	3	---	2	1	4	8	21
TOTAL	17	9	2	7	7	11	21	74

* This region's sites are operated by DEQ and NPS

Number of Criteria Pollutant Monitoring Sites

Air Quality Planning Areas for the Commonwealth of Virginia



**Alexandria and Arlington are also attainment/maintenance for the 1985 CO NAAQS.*

- Hampton Roads Attainment Area/
Voluntary Ozone Advance Action Plan**
- Gloucester County
 - Isle of Wight County
 - James City County
 - York County
 - City of Chesapeake
 - City of Hampton
 - City of Newport News
 - City of Norfolk
 - City of Poquoson
 - City of Portsmouth
 - City of Suffolk
 - City of Virginia Beach
 - City of Williamsburg

- Richmond-Petersburg Attainment Area/
Voluntary Ozone Advance Action Plan**
- Charles City County
 - Chesterfield County
 - Hanover County
 - Henrico County
 - Prince George County
 - City of Colonial Heights
 - City of Hopewell
 - City of Petersburg
 - City of Richmond

- Caroline County Attainment Area/
Voluntary Ozone Advance Action Plan**

- Fredericksburg Attainment Area/
Voluntary Ozone Advance Action Plan**
- Spotsylvania County
 - Stafford County
 - City of Fredericksburg

Updated 7/25/2017

<http://www.deq.virginia.gov/programs/air/airqualityplans/ozoneandp25regionalplanningactivities.aspx>

Appendix B

AIRSData – Access to national and state air pollution concentrations and emissions data
<https://www.epa.gov/outdoor-air-quality-data>

Air Emission Sources
<https://www.epa.gov/air-emissions-inventories/air-emissions-sources>

Air Now – Ozone mapping, AQI, and real time data
<http://www.airnow.gov>

Air Now – Air Quality Index Information
<http://www.airnow.gov/index.cfm?action=aqibasics.aqi>

American Lung Association:
<http://www.lungusa.org/>

AQS Data Mart (AQS data for the scientific and technical community):
https://aqs.epa.gov/aqsweb/documents/data_mart_welcome.html

CASTNET (Clean Air Status and Trends Network)
<http://www.epa.gov/castnet>

Department of Environmental Quality link:
<http://www.deq.virginia.gov/>

Education for teachers and children:
<http://airnow.gov/index.cfm?action=learning.forteachers>
<http://airnow.gov/index.cfm?action=student.teachers>

EPA Popular Resources
<https://www.epa.gov/learn-issues>

EPA's Technology Transfer Network (TTN) – Ambient Monitoring Technology Information Center (AMTIC)
<https://www.epa.gov/amtic>

IMPROVE
<http://vista.cira.colostate.edu/improve>

MARAMA
<http://www.marama.org/homepage>

Nonattainment area descriptions:
<https://www.epa.gov/green-book/green-book-national-area-and-county-level-multi-pollutant-information>

U.S. EPA:
<http://www.epa.gov>

2016 3-Day Monitoring Schedule for PM2.5 and 6-Day Monitoring Schedule for PM10:
<https://www3.epa.gov/ttn/amtic/calendar.html>

Code of Federal Regulations – 40 CFR 50 & 58

http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40tab_02.tpl

Virginia Ambient Air Monitoring Data Reports

<http://www.deq.virginia.gov/Programs/Air/AirMonitoring/Publications.aspx>

Air Quality System (AQS)

<https://www.epa.gov/aqs>

References